

Triterpene saponinins with oleanene skeleton: chemotypes and biological activities

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Critical survey of a selected class of pentacyclic triterpenes — the oleanane family, is presented based on current literature in order to underline their value for medicinal chemistry and drug development potential. Oleanenes may be considered as a renewable resource of valuable research materials which are structurally diverse, inherently biocompatible and have built-in affinity for many categories of functional proteins. Although availability of particular compounds from natural sources may be very low, synthetic methods elaborated by generations of chemists, secure a way to obtaining desirable structures from commercial starting materials.

Key words: pentacyclic triterpenes, oleanane derivatives, oleanolic acid exploratory chemistry and experimental pharmacology

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INTRODUCTION

Pentacyclic triterpenes (PTT) constitute a large class of natural products widespread in the Plant Kingdom (Azimova, 2013; Dinda *et al.*, 2010). Extraordinary structural diversity in this category of secondary metabolites has fascinated chemists and biologists alike for well over half a century. Chemical concept of the biogenetic “isoprene rule” as a basis of terpenoid assembly in plants, advanced in a seminal paper by Swiss chemists (Eschenmoser *et al.*, 1955) has been recapitulated upon the publication 50-thieth anniversary, with appropriate commentaries, outlining the substantial field of natural product research related to biomimetic carbocyclization which leads to triterpenoids and steroids (Eschenmoser & Arigoni, 2005). Parallel effort of biosynthetic studies afforded definite conclusion, that despite many possible pathways executed by highly specific squalene cyclases (OSC), a given PTT comprising five carbocycles and eight centers of chirality is assembled in a single enzymatic step from common linear precursor (Lodeiro *et al.*, 2007; Pollier *et al.*, 2013). Although knowledge of biosynthesis, including active sites topology of specific squalene cyclases, and structures of terpenoids in plants is presently well established, their biological functions are not entirely understood. Nevertheless, it is generally recognized that biogenesis of terpenes as such and also in their glycosylated forms (e.g. as saponins) must offer some advantage of environmental nature to the plant host, which tends to be evolutionary favored and genetically preserved. The present view of triterpene biosynthesis constitute of three distinct groups of enzymatic

transformations: 2,3-oxidosqualene cyclization, which results in formation of such triterpene skeletons as oleanane, ursane, hopane, etc. (Pollier *et al.*, 2013); action of cytochrome type multifunctional oxidases, which are responsible for introduction of hydroxyl groups and also their subsequent oxidations (Fukushima *et al.*, 2011); action of acylases and glycosyltransferases, which complete the chain of events leading to saponins which are the end products of chemical defence and allelopathic chemicals mediating interspecies interactions (Geisler *et al.*, 2013). Historically, biological activity of saponins attracted more attention than their respective triterpene aglycons (genins) (Hostettmann & Marston, 2005; Negi *et al.*, 2013). More recently, it has been realized that various natural triterpenes exhibit pleiotropic activity towards plethora of molecular targets, generating much interest of researchers and considerable potential for pharmacological studies (Sun *et al.*, 2006; Zwenger & Basu, 2008; Sheng & Sun, 2011). Yet, in general perception significance of triterpenes as prospective new drug leads is rather underscored. In order to explore and discuss their potential in some methodical way, we decided to present some structural, chemical and biological activity research data, for a restricted group of compounds, representing various chemotypes placed within strictly defined molecular framework of a particular type of triterpene skeleton originating from typical biogenetic pathway. Our choice of β -amyirin (BAR, **3**) sub-class of pentacyclic triterpenes stems from their considerable widespread occurrence and relatively good availability of pure chemical entities, which were studied as biologically active plant constituents but also gave rise to several generations of semi-synthetic derivatives bearing functionalities not found in Nature. Obviously, such choice is artificial in view of well known diversity of the squalene cyclization processes, which involves variety of OSC enzymes of different selectivity (Augustin *et al.*, 2011). Although many plants contain terpenoid secondary metabolites which result from various parallel biogenetic pathways, this review purposely focuses on only one selected type of terpenoid framework (scaffold), in order to seek and discuss pos-

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Abbreviations: API, active pharmaceutical ingredient; BAR, β -amyirin; BAC, β -amyirin cyclase; BAS, β -amyirin synthase; CYP, cytochrome P450, heme-related monooxidase; COX-2, cyclooxygenase type 2; CDDO, 2-cyano-3,12-dioxooleana-1,9(11)-dien-28-oic acid; CDDOMe, bardoxolone, 2-cyano-3,12-dioxooleana-1,9(11)-dien-28-oic acid methyl ester; DMAP, 4-dimethylaminopyridine; EDCl, 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride; GTS, glycosyltransferase; HCSE, horse chestnut seed extract; HCV, hepatitis C virus; IKK β , subunit of I κ B kinase complex; iNOS, inducible nitrogen oxide synthase; NF κ B, nuclear factor kappa B; Nrf2, nuclear protein activating Antioxidant Response Element; OSC, oxidosqualene cyclase; PTT, pentacyclic triterpenes; SAR, structure-activity relationship

sible differences in biological activities between various products of enzymatic oxidations and secondary derivatizations. Triterpenes are abundant in many dicotyledonous plants (over 80 families are involved; there are over 20000 of individual triterpenoid natural products), mainly in glycosylated form, as mono- and bis-desmosidic saponins (Azimova 2013; Dinda *et al.*, 2010) and β -amyrin can be considered the simplest, most popular aglycone of the class and a model scaffold structure for triterpene based natural and semisynthetic pharmacophores. Since 1985, newly isolated triterpenes are regularly covered in annual review initiated by the Royal Chemical Society (Hill & Connolly, 2013 and earlier annual reports).

HYDROXYLATED PENTACYCLIC TRITERPENES (PTT) DERIVED FROM OLEANYL CATION

(*S*)-2,3-Epoxy-squalene, which results from biogenetic condensation of isoprenoid phosphates, undergoes in Nature a sequence of enzymatic transformations, producing carbocationic PTT intermediates (Xue *et al.*, 2012), which can undergo Wagner-Merwein type rearrangements but are also susceptible to a spontaneous stabilization with concomitant formation of an olefinic bond as illustrated on the Scheme 1. Condensed pentacyclic framework formed, features characteristic pattern of substitution with residual 3- β hydroxyl group, double bond Δ -12,13; eight methyl groups forming two geminal arrangements (at C-4 and C-20), trans- junction of the cyclohexane ABCD rings and cis-fused rings DE. The corresponding hydrocarbon is known as olean-12-ene, which is a convenient entry for semi-systematic nomenclature for natural and synthetic derivatives. This molecular arrangement, presented on Fig. 1 with some stereochemical details elucidated from X-ray structure determination (Maartmann-Moe *et al.*, 1987; Froelich & Gzella, 2010) and key carbon atom numbering, is characterized by relatively high thermodynamic stability as evidenced by conformational energetics study conducted by EJ Corey (Surendra & Corey, 2009). Great many derivatives of β -amyrin (**3**) found in plants differ from the prototypic structure mainly by hydroxylation patterns, double bond position and oxidation level (Azimova 2013). Triterpene hydroxyl groups derivatizations found in nature are of two different kinds, stemming from acylation and/or glycosylation and their nearly combinatorial application applied by evolutionary biogenesis in plants generated many collections which can serve as individual chemotaxonomic hallmark of species. It is well established customary idea in phytochemistry, pharmacognosy and related disciplines that saponins, which are natural glycosides featuring various kinds of lipophilic steroidal or terpenoid aglycones constitute a primary subject of their study. Indeed, many triterpenes do not occur in nature in their free state, but only in form of acylated conjugates with mono or oligosaccharides. Saponins are distinct and important category of natural products, which have found their applications early in development of life sciences and their study are amply covered in the literature, from traditional to modern, which leaves them outside the scope of this review.

Our attempt to collect BAR analogs, which are scattered all over the Plant Kingdom, and present them as one structurally related chemical family, comprises compounds of natural origin sharing characteristic C₃₀ oleanane pentacyclic pattern: from primary products of OSC action, through polyhydroxylated neutral genins, to mono- and di- basic carboxylic acids. Their conven-

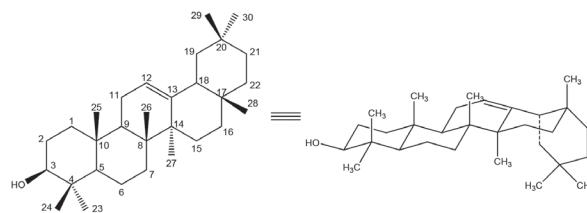


Figure 1. Conventional representations of β -amyrin structure (equivalent).

tional names usually do not reveal essential structural features; quite frequently a single PTT has several synonymic names, which can be misleading. Neutral BAR derivatives usually feature a double bond (relatively unreactive; typical olean-12-enes are reported to resist catalytic hydrogenation and exhibit atypical reactivity towards oxidative reagents), hydroxyl groups, primary and secondary, of different reactivity. Stereochemistry of secondary hydroxyl groups reflects specificity of particular hydroxylases, which are P450 cytochrome type enzymes (Geisler *et al.*, 2013) characteristic for a given plant species. Typically hydroxylated positions of oleanene skeleton, such as: 2, 3, 6, 11, 15, 16, 21 and 22 can carry hydroxyl groups of either α - or β -configuration. Oxo functions are occasionally encountered within BAR analogues, both as aldehyde or keto group. It has to be mentioned that “epoxy” prefix is applied in conventional nomenclature of PTT in two different meanings: as vicinal and non-vicinal anhydro-diol arrangement, which can be misleading. Acidic function is usually (but not always) revealed in the trivial name of a natural triterpene. Frequently encountered ending: “genin” testifies to origin of newly identified specimen at the time of its discovery as an aglycone of saponins. Apart from venerable attempts at total synthesis, chemistry of simpler PTT have not been particularly developed, thus relatively succinct descriptions of some hydroxylated oleanenes selected from collection of compounds **1–64**, presented in Table 1. This changes considerably as we enter hydroxylated acid category (Table 2), which can be explained by greater availability of substrates and wider scope of conceivable chemical transformations. Two collections of naturally occurring triterpenes are presented, to illustrate variety of chemotypes encountered within oleanene PTT family. The choice of molecular objects has been arbitrary, but according to authors intention, it reflects a level of interest measured as a number of available literature references.

β -AMYRIN AND ISOMERIC OLEANENE MONOOLS

(*S*)-2,3-Epoxy-squalene, the key intermediate in biosynthesis of higher isoprenoids, can be cyclized in several manners by plethora of specific enzymes – oxidosqualene cyclases (OSC), and among them β -amyrin synthases (BAS) are relatively numerous (Xue *et al.*, 2012). β -Amyrin (**3**, BAR, also known as β -amyrenol), occurs in free form and also as fatty acid esters in many plant resins, latexes and waxes (together with α -amyrin, which belongs to ursane class of PTT). It has been isolated, among others, from mistletoe, clove, sugar beet, and olive leaves. Following a wave of TPP total syntheses (in racemic form) started around 1970, enantioselective methods have been developed, culminating in synthesis of **3** and some analogs by EJ Corey (Surendra & Corey, 2009). The syntheses are listed among highest

Table 1. Neutral PTT genins of the oleanane class.

No	Name	Substitution pattern	CAS No	Occurrence	Literature
1	Abrisapogenol C	Olean-12-ene-3 β ,21 β ,22 β ,29-tetrol	129773-44-5	Sophora	Ikeda 2005
2	Amyrenonol	Olean-12-ene-3 β -hydroxy-11-one	38242-02-3	Buddleia	Agata 1965
3	β -Amyrin	Olean-12-ene-3 β -ol	559-70-6	widespread	David 1950
4	δ -Amyrin	Olean-13-ene-3 β -ol	508-04-3	Spartium	Musgrave 1952
5	β -Amyrone	Olean-12-ene-3-one	638-97-1	Ilex, Quercus	Ling 2010
6	Anhydrosophoradiol	Olean-12,21-diene-3 β -ol	86425-22-1	Calotropis gigantea	Kitagawa 1983
7	Armilarigenin	Olean-12-ene-3 β ,16 α ,28-trihydroxy-21-one	22570-57-6	Jacquinia	De Maheas 1969
8	Barrigenol A1	Olean-12-ene-3 β ,15 α ,16 α ,22 α ,28-pentol	15448-03-0	Barringtonia	Shamma 1962
9	Barrigenol R ₁	Olean-12-ene-3 β ,15 α ,16 α ,21 β ,22 α ,28-hexol	15399-43-6	Barringtonia, Aesculus	Errington 1967
10	Barringtogenol C (Theasapogenol B)	Olean-12-ene-3 β ,16 α ,21 β ,22 α ,28-pentol	13844-01-4 (17304-24-4)	Aesculus, Styrax, Camellia, Barringtonia	Hiller 1975
11	Barringtogenol D	Olean-12-ene-16 α ,21 α -anhydro 3 β ,22 α ,28-triol	19882-11-2	Barringtonia, Aesculus	Wulf 1969
12	Camelliagenin A (Theasapogenol D)	Olean-12-ene-3 β ,16 α ,22 α ,28-tetrol	53241-41-1	Camellia, Ternstroemia japonica	Itokawa 1969
13	Camelliagenin B	Olean-12-ene-3 β ,16 α ,22 α ,28-tetrahydroxy-23-al	14511-74-1	Camellia, Ternstroemia japonica	Yosioka 1972
14	Camelliagenin C (Theasapogenol C)	Olean-12-ene-3 β ,16 α ,22 α ,23,28-pentol	14440-27-8	Camellia	Ito 1967
15	Careyagenol D	Olean-12,15-diene-3 β ,21 α ,22 α ,28-tetrol	52591-13-6	Carea arborea	Mahato 1973
16	Careyagenol E	Oleana-11,13(18)-diene-3 β ,21 β ,22 α ,28-tetrol	55907-35-2	Carea arborea	Mahato 1974
17	Castanogenol	Olean-12-ene-2 β ,3 β ,23,28-tetrol	26553-62-8	Castanosperrum	Rao 1969
18	Castanopsol	Olean-12-ene-1 α ,3 β -diol	66088-16-2	Castanopsis	Hui 1975
19	Complogenin	Olean-12-ene-11-one-3 β ,22 β ,23 β -tiol	149471-34-1	Astragalus	Cui 1992
20	Cyclamiretin A	Olean-13,28-anhydro-3 β ,16 α -dihydroxy-29-al	5172-34-9	Cyclamen, Primula	Tschesche 1964
21	Daturadiol	Olean-12-ene-3 β ,6 β -diol	41498-79-7	Datura	Kocor 1973
22	Epigermanicol	Olean-18-ene-3 α -ol	64727-44-2 (acetate)	Euphorbia	Estrada 1957
23	Eryngiol A	Olean-12-ene-3 β ,16 α ,21 β ,22 α ,28,29-hexol	50982-44-0	Eryngium	Hieler 1973
24	Erythrodiol	Olean-12-ene-3 β ,28-diol	545-48-2	Erythroxylon, Erythrina	Manez 1997
25	Escigenin (Aescigenin)	Olean-12-ene-16 α ,21 α -anhydro-3 β ,22 α ,24,28-tetrol	17806-68-7	Aesculus, Barringtonia	Wulf 1969
26	Germanicol	Olean-18-ene-3 β -ol	465-02-1	Euphorbia	Wang 2010

No	Name	Substitution pattern	CAS No	Occurrence	Literature
27	Germanidiol	Olean-18-ene-2 α ,3 β -diol	10179-23-4	Rhododendron	Nakamura 1965
28	Gymnestrogenin	Olean-12-ene-3 β ,16 β ,21 β ,23,28-pentol	19942-02-0	Gymnema	Stocklin 1968
29	Gymnemagenin	Olean-12-ene-3 β ,16 β ,21 β ,22 α ,23,28-hexol	22467-07-8	Gymnema	Liu 1992
30	Gymnorhizol	Olean-13(18)-ene-3 α -ol	52647-56-0	Bruguera gymnorhiza	Musgrave 1952
31	Gymnosporol	Olean-12-ene-3,11-dione	29620-99-3	Gymnosporia	Govindachari 1970
32	Hirsudiol	Olean-13(18)-ene-2 α ,3 α -diol	109269-91-2	Cocculus	Ulubelen 1977
33	Isoescigenin	Oleana-12,15-diene-3 β ,21 α ,22 β ,24,28-pentol	2548-60-9	Aesculus	Thomson 1966
34	Kudzusapogenol A	Olean-12-ene-3 β ,21 β ,22 β ,24,29-pentol	96820-46-1	Sophora Puerariae	Ikeda 2005
35	Kudzusapogenol C	Olean-12-ene-3 β ,21 β ,24-triol	96820-47-2	Puerariae	Kinjo 1985
36	Longispinogenin	Olean-12-ene-3 β ,16 β ,28-triol	465-94-1	Gymnema	Ukiya 2002
37	Maniadiol	Olean-12-ene-3 β ,16 β -diol	595-17-5	Manila, Calendula	Ukiya 2002
38	Moradiol	Olean-18-ene-3 β ,28-diol	568-49-0	Planchonia, Buxus	Abramson 1973
39	Oxytrogenol	Olean-12-ene-3 β ,22 β ,24,30-tetraol	121994-07-8	Oxytropis glabra	Sun 1990
40	Pridentigenin E	Olean-12-ene-3 β ,16 α ,28,30-tetrol	3345-34-4	Lysimachia	Ahmad 1980
41	Primulagenin A	Olean-12-ene-3 β ,16 α ,28-triol	465-95-2	Primula, Jacquinia	Kitagawa 1972
42	Priverogenin A	Olean-12-ene-3 β ,16 α ,22 α -trihydroxy-28-al	18443-26-0	Primula, Lysimachia	Kitagawa 1972
43	Priverogenin B	Olean-12-ene-13,28-anhydro-3 β ,16 α ,22 α -triol	20054-97-1	Primula, Lysimachia	Kitagawa 1972
44	Protoescigenin	Olean-12-ene-3 β ,16 α ,21 β ,22 α ,24,28-hexol	20853-07-0	Aesculus	Gruza 2013
45	Pulcherol	Olean-12-en-3 α -ol	6811-63-8	Euphorbia	Castro 2013
46	Querretarol	Olean-12-ene-3 β ,28,30-triol	3767-05-3	Vernonia	Barton 1968
47	Saikogenin A	Olean-11,13-diene-3 β ,16 α ,23,28-tetraol	5092-09-1	Bupleurum	Takagi 1980
48	Saikogenin B	Olean-9(11),12-diene-3 β ,16 β ,28-triol	6002-68-2	Bupleurum	Kubota 1967
49	Saikogenin D	Olean-11,13(18)-diene-3 β ,16 α ,23,28-tetraol	5573-16-0	Bupleurum	Kubota 1967
50	Saikogenin E	Olean-11-ene-13,28-anhydro-3 β ,16 β -diol	79786-12-2	Bupleurum	Takagi 1980
51	Saikogenin F	Olean-11-ene-13,28-anhydro-3 β ,16 β ,23-triol	14356-59-3	Bupleurum	Takagi 1980
52	Saikogenin G	Olean-11-ene-13,28-anhydro-3 β ,16 α ,23-triol	18175-79-6	Bupleurum	Takagi 1980
53	Saikogenin H	Olean-9(11),12-diene-3 β ,16 β ,23,28-tetraol	99365-24-9	Bupleurum	Shimizu 1985

No	Name	Substitution pattern	CAS No	Occurrence	Literature
54	Saikosapogenin B ₁	Olean-11,13-diene-3 β ,16 β ,24,28-tetrol	58558-08-0	Bupleurum	Takagi 1980
55	Saikosapogenin B ₂	Olean-11,13-diene-3 β ,16 α ,24,28-tetrol	58316-41-9	Bupleurum	Takagi 1980
56	Sophoradiol	Olean-12-ene-3 β ,22 β -diol	6822-47-5	Sophora	Kinjo 2003
57	Soyasapogenol A	Olean-12-ene-3 β ,21 β ,22 β ,24-tetraol	508-01-0	Glycine max	Smith 1958
58	Soyasapogenol B	Olean-12-ene-3 β ,22 β ,24-triol	595-15-3	Glycine max, Astragalus	Kinjo 2003
59	Soyasapogenol C	Olean-12,21-diene-3 β ,24-diol	595-14-2	Glycine max, Trifolium	Cainelli 1958
60	Soyasapogenol E	Olean-12-ene-3 β ,24-dihydroxy-22-one	6750-59-0	Anthyllis vulneraria	Nartowska 2001
61	Tangjinol	Olean-12-ene-3 β ,6 β ,7 β ,16 β ,23,28-hexaol	21963-76-8	Barringtonia	Row 1963
62	Taraxerane	Olean-14(15)-ene-3 β -ol	127-22-0	Pouteria caimito	Ardon 1973
63	Theasapogenol A	Olean-12-ene-3 β ,16 α ,21 β ,22 α ,23,28-hexol	13844-22-9	Camellia sasanqua	Yosioka 1972
64	Theasapogenol E (Camelliagenin E)	Olean-12-ene-23-al-3 β ,16 α ,21 β ,22 α ,28-pentol	15399-41-4	Camellia sasanqua	Yosioka 1972

achievements of the art of academic molecular design and assembly, but having no practical value, they did not influence availability of PTT materials. Although it is known that in plants, the primary cyclization products of OSC are in turn a subject to subsequent oxidative transformations, which are performed by CYP type oxidases (Pollier *et al.*, 2013; Geisler *et al.*, 2013) the prospects of exploiting this knowledge in design of biotechnological processes (Moses *et al.*, 2013) seems rather distant.

Majority of PTT end up with β -3-OH group but secondary metabolic transformations can change it. Thus, β -amyrin corresponding 3-epi compound, called Pulcherrol has been found in *Euphorbia pulcherrima* and *Eupatorium bavanense*. Their oxidation product: β -Amyrone (Pulcherrone) has been isolated from several plant sources. Isomeric 3-monoalcohols featuring Δ -13,18 unsaturation are known under names: δ -amyrin (3 β -) and gymnorrhizol (3 α -), while Δ -18 analogs are called germanicol and epi-germanicol, respectively. Together with dienols, there are well over a dozen of single oxygen derivatives of pentacyclic oleanane hydrocarbon. Naturally, number of derivatives grows very quickly as consecutive substituents are introduced. Compounds 1–64 in Table 1 represent only a small fraction of known non-acidic derivatives of β -amyrin. They are collected together against phytochemical tradition pooling secondary metabolites of a particular plant, in order to demonstrate functional and stereochemical diversity within a single class of PTT, associated with one particular type of terpenoid skeleton. Some of the presented compounds are considerably more available than other, which is reflected in number of studies, going beyond chemical structure elucidation. The following short notes are intended to bring attention to particular PTTs, which show promise as biologically active compounds, either in terms of selectivity or efficacy.

NONACIDIC OLEANENES WITH MULTIPLE FUNCTIONALITIES

In comparison to some other triterpenes, like for example betulin, which belong to lupine class, and evoked enormous interest both: as biologically active compound in its own right, and also as a starting point for synthetic exploration towards better drug lead compounds, non-acidic oleanenes are much less developed as prospective pharmacophores. Only very recently several review papers brought attention to their potential (Salminen *et al.*, 2008; Yadav *et al.*, 2010; Podolak *et al.*, 2010; Thoppil & Bishayee, 2011; Liang *et al.*, 2011; Bishayee *et al.*, 2011; Parmar *et al.*, 2013; Yin, 2012). These references concern PTT belonging to all structural types and concentrate on the most popular compounds, with relatively good availability. Yet, despite of growing interest, most PTT compounds from this category, like majority of compounds listed in Table 1, remain in obscurity as far as modern pharmacological research is concerned.

Erythrodiol

Erythrodiol (olean-12-ene-3 β ,28-diol, Homoolestranol, 24), is present in olive pressing residues as well as many other plants, including such important agricultural crop as soy. Despite widespread occurrence, chemical synthesis by one step reduction from easily available of 105 esters is a viable alternative to isolation. Its occurrence in olive oil rose some food safety concerns, soon dropped after toxicology examination. The therapeutic efficiency of erythrodiol on different experimental models of in-

flammation has been reported (Manez *et al.*, 1997; de la Puerta *et al.*, 2000). Other biological activities discussed, include anticancer (Nishino *et al.*, 1988) and antihypertensive effects (Rodríguez-Rodríguez *et al.*, 2004).

Longispinogenin

Longispinogenin (olean-12-ene-3 β ,16 β ,28-triol, **36**), is a constituent of flower extract of chrysanthemum (*Chrysanthemum morifolium*). It exhibits inhibitory effects on Epstein–Barr virus early antigen (EBV-EA) activation induced by the tumor promoter (Ukiya *et al.*, 2002). The inhibitory effects of this compound were almost equivalent to or stronger than that of glycyrrhetic acid, which is a potent antitumor promoter (Konoshima *et al.*, 1999).

Soyasapogenols

Soyasapogenols have been relatively well studied because their source — soybeans (*Glycine max* Merrill) constitute one of the most important agricultural crop and basis for food technology in the global scale. In the past, soy protein used to contain several non-nutrient impurities, like phytic acid derivatives, isoflavones and saponins, all considered undesirable but difficult to remove because of similar physicochemical properties. Isoflavones recoverable from soy technology waste materials, have made their way to innumerable food supplements as phytoestrogens. To our knowledge soy saponins (and their corresponding genins) have not yet surfaced as commercial products (Zhang & Popovich, 2009) but their potential should not be underestimated. Among PTT genins derived from soy, the main constituent soyasapogenol B: olean-12-ene-3 β ,22 β ,24-triol (**58**), stands out as the compound with pronounced antiviral (HSV-1) activity (Ikeda *et al.*, 2005). Interestingly, corresponding tetraol — soyasapogenol A, is much less active, which contradicts simple idea that more functionality renders better efficacy.

Kudzusapogenols

Kudzusapogenol A (olean-12-ene-3 β ,21 β ,22 β ,24,29-pentol) is the principle genin of *Pueraria lobata* root saponins but closely related kudzusapogenols B and C are also known. Their structure is closely related to genins of soya and horse chestnut, featuring C-24 primary hydroxyl group. The plant has rather special status in traditional Chinese medicine (TCM) as an agent against vertigo, headache and migraine. Its present applications against alcohol dependence are connected with the presence of isoflavone — puearrine. The genins have been examined for antiviral activity and shown to perform similarly to soyasapogenols.

Barringtogenols

Barringtogenol C (olean-12-ene-3 β ,16 α ,21 β ,22 α ,28-pentol, **10**) and barringtogenol D (olean-12-ene-16 α ,21 α -epoxy-3 β ,22 α ,28-triol, **11**) unlike many other saponins constitute triterpenoid framework for saponins found in numerous plant sources, leaves of *Careya arborea*, seeds of *Aesculus hippocastanum* L., seeds of *Barringtonia acutangula* or fruits of *Styrax japonica*. As a result, barringtogenol C is known under several names, like: Theasapogenol B, Jegosapogenol A, Saniculagenin D, Careyagenol A, Giganteumgenin M, Aescinidin, Acutangenol B. Although biological activities are much better recognized for their corresponding saponins, e.g. antiprotozoal activity against *Leishmania donovani* (Mandal *et al.*, 2006), some prospects

suggested for their antidiabetic applications warrant further interest and research (Yoshikawa *et al.*, 1996).

Gymnemagenins

Among treasures of Ayurvedic medicine, gurmar (in Hindi: sugar destroyer), referring to *Asclepiadaceae* family herb — *Gymnema sylvestre* R. Br., is one of the best recognized by Western modern pharmacognosy. Consumption of its leaves has profound effect on taste — it selectively suppresses response to sweeteners like sucrose, saccharin and cyclamate. Herbal preparations and plant extracts have been tested for variety of health supporting activities (anti-inflammatory, antibacterial, antiobesity, hypolipidemic) with positive results and gymnemic acids — glucuronides of partially acylated PTT — gymnemagenin have been found responsible for potent antidiabetic activity (Sanjea *et al.*, 2010; Patel *et al.*, 2012). A sensitive HPLC-tandem MS analytical method for determination of gymnemagenin in rat plasma made it possible to monitor active principle dosing and to study pharmacokinetics (PK), following administration of variety *Gymnema* preparations (Kamble *et al.*, 2013), since direct determination of gymnemic acids does not seem feasible. Careful chemical investigation of *Gymnema* constituents revealed very complex picture, in which oleanane glycosides: gymnemic acids, gymneasins and gymneasides, coexist with dammarane-type saponins, also named gymneasides (Porchezian & Dobiryal, 2003). Many patents have been filed for health care applications of *Gymnema* where genins are claimed for beneficial effect, along with their glycosides.

Theasapogenins

Tea leaves processing gives opportunity to isolate saponin concentrates which are rich in PTTs. These material attained position of commodities in food industry and applications in other technical capacities, like cosmetics, paints and plastics related processes. In this respect, their situation is much like for soyasaponins whereas scale effect exert pressure for applications. Thus far theasapogenins, and accompanying cammeliasapogenins are not yet available as chemically certified materials. Nevertheless, potential of theasapogenols A, B, E and cameliagenins A and D as highly functionalized oleanenes and prospective molecular probes is obvious (Morikawa *et al.*, 2006).

Escigenins

Aesculus hippocastanum L., decorative tree widespread throughout of moderate and sub-tropical climat zone has been known for curative properties of its seed extract (HSE) since the time immemorial. Ethnopharmacological tradition of HSE application for prevention and treatment of vascular ailments stand well against modern clinical criteria. Contemporary phytochemical studies list extensive group of saponins derived from partly acylated PTT compounds — all escins, isoescins and aesculosides (isolated from *A. chinensis* Bunge), can be attributed to four genins: protoescigenin, barringtogenol C, barringtogenol D and escigenin. Although HSE is an active principle of numerous herbal preparations, neither saponins nor saponins are available as individual chemicals (with exemption of couple phytochemical standards sold in miligram quantities). Recent elaboration of a process for isolation and purification of protoescigenin **44** affords new opportunity for research in this area (Gruza *et al.*, 2013).

CARBOXYLIC ACIDS FROM OLEANENE GROUP

There are eight methyl substituents present in most PTT belonging to oleanane class and they can be subject of enzymatic oxidation, resulting in primary alcohols, corresponding aldehydes and carboxylic acids but phytochemical studies reveal that probability of such biotransformation vary greatly for particular methyl positions. PTT natural products bearing carboxylic functions are widespread, particularly as genins of bisdesmosidic saponins which are characterized by unique structural feature — combining in one molecule two oligosaccharide chains with distinctly different types of linkage: one glycosidic, typically placed in position 3 and one glycosyl ester involving terpene carboxylic group and anomeric position of an oligosaccharide. Since acidic PTT are relatively easily separable from the bulk of neutral secondary metabolites they became a forefront materials for various technical applications and became subjects of countless patents, which are not listed. Acidic PTT (in some cases known only as genins of particular saponins) are collected in Table 2.

Oleanolic acid

Olean-12-ene-3 β -hydroxy-28-oic acid (oleanolic acid, **105**) is one of the most widespread secondary metabolites (Pollier & Goossens, 2012) — it has been isolated from more than 1620 dicotyledone plant species! Its synonyms include: Astriantiagenin C, Caryophyllin, Giganteumgenin C, Gledigenin 1 and Virgaureagenin B, which clearly indicates its presence in various saponins. Its occurrence in unconjugated form is often observed in plant waxes serving as physical barrier to pathogens attack and preventing water loss. First isolated over a century ago, it has a long story of chemical investigation but its high resolution X-ray structure was determined only recently (Froelich & Gzella, 2010) Plant originated **105** is a result of common cyclase activity (BAC), followed by three step oxidation of **3**, carried out sequentially on C-28 methyl group by cytochrome enzyme classified as CYP716A12 (Scheme 1 and Scheme 2). Many medicinal plants, such as *Panax ginseng*; *Hedera helix*; *Calendula officinalis*; *Thymus vulgaris*; *Rosmarinus officinalis*, *Viscum album*, etc., have been shown to contain oleanolic acid, often together with other PTT components. Presently, the principal source of **105**, turned commercial, are olive oil manufacturing process waste materials. Olive tree (*Olea europaea*, Oleaceae) leaves, collected together with olives during harvest contain about 3% of unconjugated oleanolic acid. Considering large scale of olive oil manufacturing (ca. 4 mln metric tonnes per annum) this secondary metabolite could be considered marketable specialty chemical. Unfortunately, its pricing as a chemical reagent does not favor widespread research. Thus far, **105** has a status of active pharmaceutical ingredient (API) only in China, but it is also present in innumerable OTC (over the counter) multicomponent herbal preparations aimed at pharmaceutical, nutritional and cosmetic markets, globally. In line with predicted growing demand for the compound, there are also ongoing R&D projects towards novel biotechnological processes utilizing cloned or expressed in bacteria enzymes from triterpene biosynthetic cascade, targeting **105** as well as other PTT secondary metabolites.

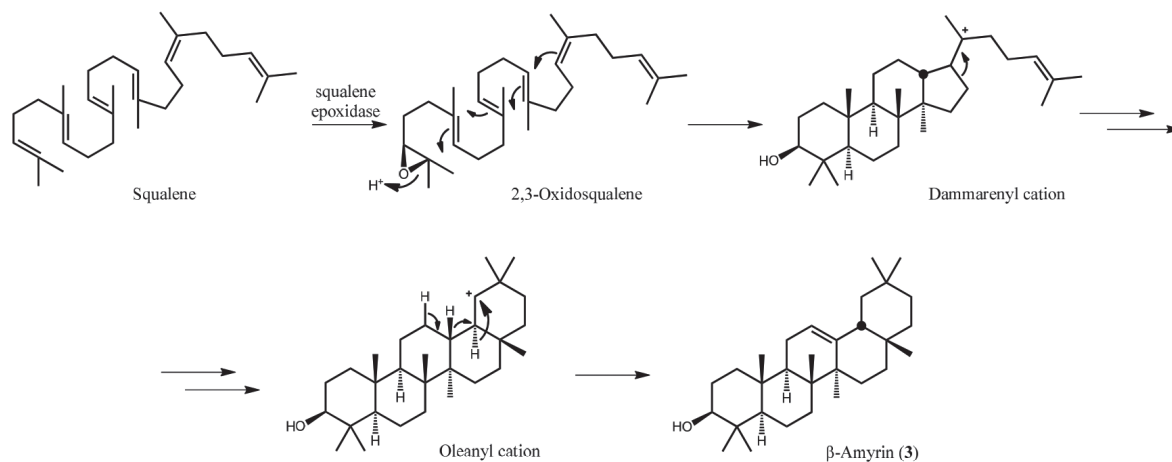
A list of pharmacological activity of **105** is extensive. It is an antioxidant in more than one sense: as a free radical scavenger but also as an inducer of the Nfr2 mediated expression of catalase and glutathione synthase. Hepatoprotect-

Table 2. Acidic PTT saponin from oleanane class.

No	Name	Substitution pattern	CAS No	Occurrence	Literature
65	Acacic A	Olean-12-ene-3 β ,16 α ,21 β -trihydroxy-28-oic acid	1962-14-7	Acacia, Entada	Barua 1977
66	Acinosolic A	Olean-12-ene-2 β ,3 β -dihydroxy-28,30-dioic acid	95260-97-2	Phytolacca	Glombitza 1975
67	Acutangulic A	Olean-12-ene-2 α ,3 β ,18 β -trihydroxy-28-oic acid	60369-84-8	Barringtonia	Anjaneyulu 1978
68	Albigenic A	Olean-13(18)-ene-3 β ,16 α -dihydroxy-28-oic acid	664-40-4	Albizzia, Helianthus	Barua 1959
69	Amooranin	Olean-12-ene-25-hydroxy-3-oxo-28-oic acid	175096-94-3	Amoora rohituka	Rabi 2002
70	Arjungenin	Olean-12-ene-2 α ,3 β ,19 α ,23-tetrahydroxy-28-oic acid	58880-25-4	Terminalia	Row 1962
71	Arjunic A	Olean-12-ene-2 α ,3 β ,19 α -trihydroxy-28-oic acid	31298-06-3	Terminalia	Row 1970
72	Arjunolic A	Olean-12-ene-2 α ,3 β ,23-trihydroxy-28-oic acid	465-00-9	Terminalia arjuna	Row 1962
73	Augustic A	Olean-12-ene-2 β ,3 β -dihydroxy-28-oic acid	26707-60-8	Perilla, Medicago	Banno 2004
74	Azizic A	Olean-12-ene-3 β ,6 α -dihydroxy-27,28-dioic acid	72959-99-0	Cornulaca	Dawidar 1979
75	Barringtogenic A	Olean-12-ene-2 α ,3 β -dihydroxy-23,28-dioic acid	471-58-9	Barringtonia	Thomas 1960
76	Bassic A	Olean-5,12-diene-2 β ,3 β ,23-trihydroxy-28-oic acid	465-01-0	Madhuca	King 1961
77	Belleric A	Olean-12-ene-2 α ,3 β ,23,24-tetrahydroxy-28-oic acid	116787-93-0	Terminalia	Ageta 1988

No	Name	Substitution pattern	CAS No	Occurrence	Literature
78	Bayogenin	Olean-12-ene-2 β ,3 β ,23-trihydroxy-28-oic acid	6989-24-8	Solidago spp, Castanospermum	King 1954
79	α -Boswellic A	Olean-12-ene-3 α -hydroxy-24-oic acid	471-66-9	Boswellia	Beaton 1956
80	Caulophylogenin	Olean-12-ene-3 β ,16 α ,23-trihydroxy-28-oic acid	52936-64-8	Caulophyllum	Strigina 1974
81	Cincholic A	Olean-12-ene-3 β -hydroxy-27,28-dioic acid	5948-32-3	Cinchona	Tschesche 1963
82	Crategolic A	Olean-12-ene-2 α ,3 β -dihydroxy-28-oic acid	4373-41-5	Crataegus	Caglioti 1961
83	Dianic A	Olean-12-ene-3 β ,29-dihydroxy-23,28-dioic acid	91652-29-8	Dianthus	Oshima 1984
84	Echinocystic A	Olean-12-ene-3 β ,16 α -dihydroxy-28-oic acid	510-30-5	Echinocystis, Albizzia	Frazier 1944
85	Entagenic A	Olean-12-ene-3 β ,15 α ,16 α -trihydroxy-28-oic acid	5951-41-7	Entada	Barua 1983
86	Esculentic	Olean-12-ene-3 β ,23-dihydroxy-28,30-dioic acid	56283-68-2	Phytolacca	Johnson 1974
87	Glabric A	Olean-12-ene-3 β ,21 α -dihydroxy-11-oxo-29-oic acid	22327-86-2	Glycyrrhiza	Beaton 1956
88	Glycyrrhetic A 18 β	Olean-12-ene-3 β -hydroxy-11-oxo-30-oic acid	471-53-4	Glycyrrhiza	Beaton 1955
89	Glycyrrhetic A 18 α	Olean-12-ene-3 β -hydroxy-11-oxo-30-oic acid	1449-05-4	Glycyrrhiza	Sabbioni 2005
90	Gypsogenic A	Olean-12-ene-3 β -hydroxy-23,28-dioic acid	5143-05-5	Gardenia, Euphorbia	Belous 1967
91	Gypsogenicin	Olean-12-ene-3 β -hydroxy-23-oxo-28-dioic acid	639-14-5	Saponaria	Ruzicka 1937
92	Hederagenin	Olean-12-ene-3 β ,23-dihydroxy-28-oic acid	465-99-6	Hedera, Akebia, Sapindus, Astrania	Power 1913
93	Jaligonic A	Olean-12-ene-2 β ,3 β ,23-trihydroxy-28,30-dioic acid	51776-39-7	Phytolacca	Stout 1964
94	Jacquionic A	Olean-12-ene-13 β ,28-anhydro-16 α -hydroxy-3-oxo-30-oic acid	97557-53-4	Jacquinia	Hahn 1965
95	Karachic A	Olean-12-ene-3 β ,6 α -dihydroxy-28-oic acid	56119-15-4	Betula	Djerassi 1955
96	Katonic A	Olean-12-ene-3 α -hydroxy-29-oic acid	6894-46-8	Sandoricum indicum	King 1960
97	Kudzusapogenol B	Olean-12-ene-3 β ,21 β ,22 β ,24-tetrahydroxy-29-oic acid	96820-57-4	Puerariae	Kinjo 1985
98	Liquiritic A	Olean-12-ene-3 β -hydroxy-11-oxo-29-oic acid	10379-72-3	Glycyrrhiza	Canonica 1966
99	Liquiridolic A	Olean-12-ene-3 β ,21 α ,24-trihydroxy-28-oic acid	20528-70-5	Glycyrrhiza	Canonica 1968
100	Liquoric A	Olean-12-ene-16 α ,21 α -anhydro-3 β -hydroxy-11-oxo-30-oic acid	3808-79-5	Glycyrrhiza	Eigamal 1965
101	Macedonic A	Olean-11,13(18)-diene-3 β ,21 α -dihydroxy-29-oic acid	39022-00-9	Brucea, Glycyrrhiza	Kiryalov 1963
102	Maslinic A	Olean-12-ene-2 α ,3 β -dihydroxy-28-oic acid	4373-41-5	Olea europaea	Sánchez-González 2013

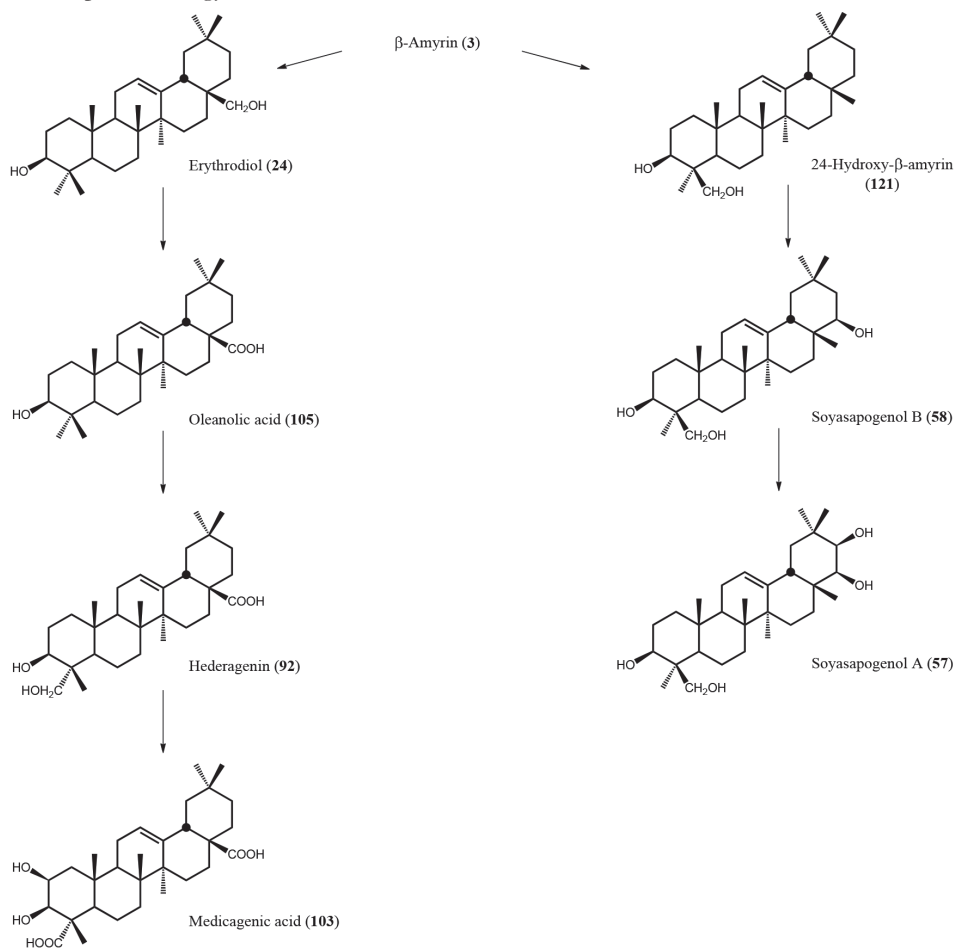
No	Name	Substitution pattern	CAS No	Occurrence	Literature
103	Medicagenic A	Olean-12-ene-2 β ,3 β -dihydroxy-23,28-dioic acid	599-07-5	Medicago	Anantaraman 1956
104	Morolic A	Olean-18-ene-3 β -hydroxy-28-oic acid	559-68-2	Agauria, Eucalyptus	Barfon 1951
105	Oleanolic A	Olean-12-ene-3 β -hydroxy-28-oic acid	508-02-1	Very widely distributed aglycone	Bishoff 1949
106	Oleanonic A	Olean-12-ene-3-oxo-28-oic acid	17990-42-0	Hedyotis lawsonii	Drefahl 1960
107	Oxyallobetulin	Olean-3 β -hydroxy-19 β ,28-olide	24035-70-9	Diospyros montana	Yoshihira 1971
108	Platycodigenin	Olean-12-ene-2 β ,3 β ,16 α ,23,24-pentahydroxy-28-oic acid	22327-82-8	Platycodon	Akiyama 1972
109	Platycogenic A	Olean-12-ene-2 β ,3 β ,16 α ,23-tetrahydroxy-24,28-dioic acid	26121-79-9	Platycodon	Kubota 1969
110	Polygalacic A	Olean-12-ene-2 β ,3 β ,16 α ,23-tetrahydroxy-28-oic acid	22338-71-2	Polygala, Solidago	Akiyama 1968
111	Presenegenin	Olean-12-ene-2 β ,3 β ,27-trihydroxy-23,28-dioic acid	2163-40-8	Polygala	Shimizu 1966
112	Pridentigenin E	Olean-12-ene-3 β ,16 α ,28,30-tetrahydroxy-28-oic acid	3345-34-4	Primula	Ito 1969
113	Quillaic A	Olean-12-ene-3 β ,16 α -dihydroxy-23-oxo-28-oic acid	631-01-6	Quillaia	Bilham 1940
114	Salviolide	Olean-3 β -hydroxy-12 β ,28-olide	76938-50-6	Salvia mexicana	Collera 1980
115	Spathodic A	Olean-12-ene-3 β ,19 α ,24-trihydroxy-28-oic acid	132194-34-4	Spathodea campanulata	Ngouela 1990
116	Tangulic A	Olean-12-ene-2 α ,3 β ,18 β -trihydroxy-23,28-dioic acid	71841-38-8	Barringtonia	Row 1978
117	Terminoic A	Olean-12-ene-2 α ,3 β ,19 α -trihydroxy-29-dioic acid	88478-13-1	Terminalia	Ahmad 1983
118	Terminolic A	Olean-12-ene-2 α ,3 β ,6 β ,23-tetrahydroxy-28-oic acid	564-13-6	Myrtus, Beclium	King 1955 and 1956
119	Virgatic A	Olean-12-ene-3 β -trihydroxy-1-oxo-28-oic acid	14356-51-5	Salvia	Ulubelen 1976
120	Zanhic A	Olean-12-ene-2 β ,3 β ,16 α -trihydroxy-23,28-dioic acid	84161-89-7	Zanha, Herniarnia	Klein 1982



Scheme 1. The key intermediates of the squalene cyclization to β -amyrin.

tive effects of the acid was demonstrated to operate in case of acute poisonings as well as in chronic diseases (Pollier & Goossens, 2012). Another biological activity proven on molecular level are interaction with farnesyl receptor FXR and suppression of some oxidative enzymes from CYP family. At the same time, the compound has been proven to be a potent inducer of the metabolic phase 2 response. Detailed structure-activity relationship, which lists effects of **105** in molecular and cellular pharmacology tests, was reviewed for

oleanane and ursane triterpenoids (Sun *et al.*, 2006). Additionally, anti-inflammatory, anti-hyperlipidemic, anti-HIV and anticancer activities of **105** have been reported (Zeng *et al.*, 2012). Observed modulation of immune-inflammatory markers by the compound suggests therapeutic implications for multiple sclerosis (Martin *et al.*, 2012). Obviously, such a collection of biological properties, potentially advantageous in prophylaxis and clinic, warrant further research on **105** and its new derivatives.



Scheme 2. Two lines of oxidative transformations of β -amyrin, carried out by CYP716A12 and CYP93E2, respectively.

Glycyrrhetic acid

Root of the herb liquorice (*Glycyrrhiza glabra* L.; Fabaceae) is known for ages as a source of sweetening agent — glycyrrhizic acid, di-glucuronide of PITT saponin: glycyrrhetic acid (**88**, also called enoxolone), which can be easily obtained in pure state by hydrolysis of liquorice saponins. Salts of **88** with alkaline metals are industrial raw materials, manufactured on thousand tons scale for use in food industry. Diammonium glycyrrhizinate is an anti-hepatic drug and the saponin similarly has strong pharmaceutical connotations. Under name: enoxolone it is known as an antiinflammatory remedy for topical use and disodium salt of its hemisuccinate, carbenoxolone has been developed in Great Britain as a drug to treat peptic ulcer (Farina *et al.*, 1998). It is speculated that anti-inflammatory action of β -glycyrrhetic acid (**88**) is caused by inhibition of steroidal of 11β -hydroxylase (Kroes *et al.*, 1997). Of increasing interests are also semi-synthetic derivatives of **88** with antitrombotic activity (Graebin *et al.*, 2010). New generation of sweeteners emerged from replacement of hydroxyl function with amine group (Ijichi *et al.*, 2005). Since the native liquorice saponins perform well in studies of supramolecular effects leading to new pharmaceutical formulations, it has been postulated that **88** could also be used for such purposes. It has to be mentioned that position of conjugated unsaturation in glycyrrhetic acid enables chemical isomerization of the ring junction D/E — thus glycyrrhetic is known in two diastereoisomeric forms: 18β - (**88**) and 18α - (**89**). Following successful entry of **102** derivative CDDO into experimental pharmacology other acidic saponins have been engaged in exploratory chemistry along parallel lines. Thus **88**, arjunolic acid **71** and α -boswellic acid **79** (usually used in a native mixture with β -regioisomer, which is ursane analog) have been converted into unsaturated cyanoketones, as prospective new antiinflammatory agents (Subba Rao *et al.*, 2008; Chadalapaka *et al.*, 2008).

Quillaic acid

The bark of Chilean soap tree (*Quillaja saponaria* Molina) has been a source of commercial saponin material since more than a half century. Although the main interest in Quillaja commodity resides in its application in food industry as detergent, foaming and/or wetting agent and emulsifier, some of the individual saponins have been found to exert potent immunoenhancing activity (Sun *et al.*, 2009), which started scientific investigation of both: saponins and their common saponin — quillaic acid (**113**). The main difference between **113** and other acidic saponins presented here resides in intermediate oxidation level of C-23 — the aldehyde function placed there offers unique opportunity to derivatize or conjugate the triterpene moiety by chemistry unavailable for more frequently encountered $-\text{OH}$ or $-\text{COOH}$ functional groups. Although **113** and its esters have been tested for antinociceptive activity (Arrau 2011), its primary application in research remains to be that of a glycoside acceptor in syntheses of bis-desmosidic saponins with immunoadjuvant function (Adams *et al.*, 2010; Ragupathi *et al.*, 2011).

Hederagenin

Hederagenin (**92**) is a common saponin for terpenoid glycosides found, among others, in common ivy (*Hedera helix*) and edible *Cenopodium quinoa*. As a component of fructus *Akebiae* extract, widely used in traditional

Chinese medicine, it has been recognized as a potent anti-depressant. Hederagenin is able to exert neuropharmacological activity by influencing serotonin and dopamine transport. It has been shown to reduce stress signs as effectively as citalopram, a proven antidepressant (Zhou *et al.*, 2010). Analytical methods for determination of **92** in body fluids were developed, following suggestion that it might be an active metabolite of native saponins (Yang *et al.*, 2011).

Maslinic acid

Maslinic acid (**102**) is a close structural analog of OLA (it differs only with one additional hydroxyl group, placed at 2α -) and likewise is a secondary metabolite of olive tree (*Olea europaea* L.). Since it occurs chiefly in the fruits, and its content in olive oil ranges from about 300 to 1300 mg/kg, its significance in human nutrition is considerable. The compound has been examined in rodent model for possible harmful effects of large doses, in hematology, clinical biochemistry and histopathology examinations, with negative results (Sanchez-Gonzales *et al.*, 2013). In a rodent study of intervention in diet-induced hyperlipidemia, both: **102** and **105** were shown to exert favorable modulations in gene expression and inhibition of the intestinal absorption and storage of cholesterol (Liu *et al.*, 2007). C-2 epimer of the maslinic acid is called augustic acid (**73**) and it is important saponin in its own right. In a program striving for PITT compounds with anti-HIV activity, **102** was coupled by an amide bond with several amino acids and peptides (Parra *et al.*, 2010). It has been demonstrated that some derivatives obtained in this way exhibited distinct activity, associated with two particular steps of the virus life cycle: entry and maturation.

Arjunolic acid and Bayogenin

Arjunolic acid (**72**) and bayogenin (**78**) represent another pair of C-2 epimeric saponins, both occurring in variety of plants, and both having significant record of use in ethnomedicine. Bark of the Indian tree from Combretaceae family — *Terminalia arjuna* is recognized as a source of **72** (a laboratory isolation process affords 210 mg from 2 kg bark; (Hemalatha *et al.*, 2010) and at the same time it is indicated as cardiac tonic with additional antilipidemic, antiinflammatory, antioxidant, and immunomodulatory properties, which evoked vivid interest in its molecular pharmacology (Gosh & Sil, 2013). While Indian researchers preferred to work on the samples of natural origin, Chinese scientists elaborated efficient synthesis of the title compounds from OLA (11 steps, about 10% yield) (Wen *et al.*, 2010). The key step involved Baldwin's cyclopalladation reaction and resulted in conversion of C-24 methyl group into acetoxymethyl substituent (Baldwin *et al.*, 1985; Neufeldt & Sanford, 2010). The semisynthetic compounds thus obtained have been examined for inhibitory activity against glycogen phosphorylase. Contrary to the initial expectations, introduction of either C-23 or C-24 hydroxyl group resulted in the loss of inhibitory potency. Remarkably, arjunolic acid **72** has been presented in chemical literature as a novel, renewable material with prospective applications in supramolecular chemistry and nanoscience (Bag *et al.*, 2008). Structural features like a rigid pentacyclic backbone, extending over a distance 1.32 nm with hydroxyl and carboxylic groups at the opposite ends have been underlined as a suitable platform for development of various chemical constructs with programmed functions.

PENTACYCLIC TRITERPENES AS AN INSPIRATION FOR CHEMISTRY AND PHARMACOLOGY

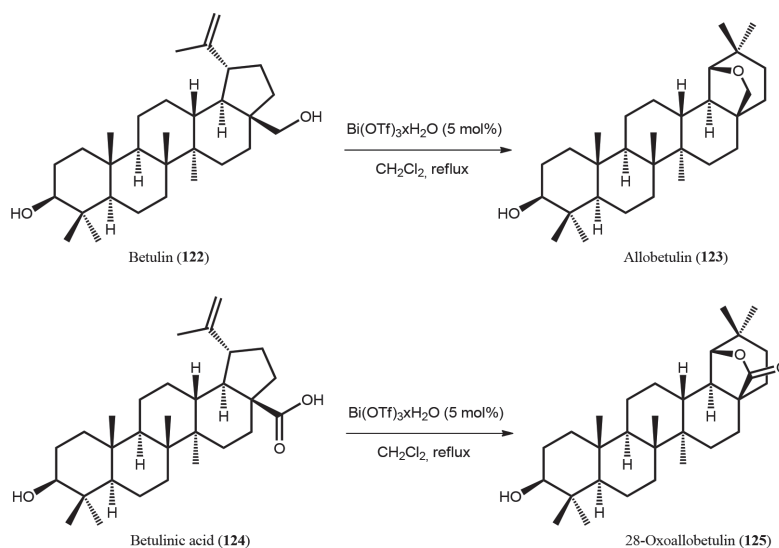
Western ethnopharmacology contains numerous examples of PTT rich plants being used as remedies for acute as well as chronic ailments, which are today classified as inflammation related pathologies. Many more such examples can be cited from Ayurvedic and traditional Chinese medicine. Recent advances in separation techniques and structural analysis helped to realize that traditional pharmacognosy, concentrating on saponins as the main active principles containing PTT, does not offer a full picture of terpenoid plant secondary metabolite potential. Although some multicomponent saponine concentrates are still used as active pharmaceutical ingredients (API; a notable example of escines from *Aesculus hippocastanus* L.), more attention is paid to individual sapogenins, particularly these available in high chemical purity. Contemporary academic chemical synthesis is a combination of sophisticated art and high technology, for which there is practically no limit within low molecular weight natural products. Thus, principal PTT, such as **3**, **102**, **105** etc., have been synthesized by total synthesis, both in racemic and enantiomeric variants (Surendra & Corey, 2009). Hydroxylated oleanene carboxylic acids, which became active principles of herbal preparations from dietary supplements category, are also starting materials for synthesis of new analogs, which reveal high potency in many modern pharmacological tests. Thus semisynthetic derivatives of **102** described in more detail below (CDDO), suppress action of inducible enzymes iNOS and COX-2, which renders them valuable tools to study anti-inflammatory effects (Liu 2005). The same group of derivatives induce apoptosis of cancerous cells, in acute myelogenous leukemia, among others (Konopleva *et al.*, 2004).

SEMISYNTHETIC DERIVATIVES OF OLEANENE

Considerable heterogeneity of such enzymes as squalene epoxide cyclases (OSC), heme related monooxidases (P450) and glycosyltransferases (GTS), which take place in multistep processes completing assembly of natural terpenoid glycosides, render them particularly difficult objects of study because as plant secondary metabo-

lites they tend to cluster as multicomponent mixtures of compounds with very similar physicochemical properties. Both: saponins and sapogenins are as a rule devoid of chromophoric groups, therefore are very difficult to separate by classical analytical methods, including chromatography using UV detection. Complex polycyclic structures of PTT with multiple centers of chirality, naturally attracted attention of synthetic chemists as an exceptional challenge, fit to test most sophisticated methods available. For decades leading academic laboratories perfected their total synthetic approaches to natural triterpene frameworks. The results of the endeavor spanned from syntheses of racemic germanicol, β -amyrin, δ -amyrin and lupeol during 1970-ties (Ireland *et al.*, 1973; Johnson *et al.*, 1993) to successful design and execution of highly enantioselective assembly of several oleanenes and lupeol by E.J. Corey (Surendra & Corey 2009) nearly 40 year later. However, these great accomplishments in the art of organic synthesis have not attained a level of practical applicability. Above quoted syntheses well exceeded 30 sequential steps and overall yields were usually much lower than 1%. Similarly, despite of great advances in plant secondary metabolites biosynthesis and biotechnology, there are no processes other than agriculture follow up isolation, available for manufacturing triterpenoids. Therefore, growing demand for pure chemical compounds from PTT category has to be satisfied by semi-synthesis, and not surprisingly carboxylic acids of natural origin stand out as suitable raw materials, since they are easier to isolate and refine to purity than neutral sapogenins. It should be mentioned that carbocation rearrangement concept, important for rationalizing diverse biomimetic cyclization pathways, has found some practical application. Thus, lupane — oleanane transformation based on Wagner-Meerwein rearrangement turned out synthetically useful reaction in case of both: betuline and betulinic acid, which are easily available starting materials (Salvador *et al.*, 2012) (Scheme 3).

Short review of oleanolic acid chemistry presented below is intended to serve as an illustration of useful interconversions of typical natural oleanene functionality into new, more pharmacophoric modes of substitution. The three functional groups of **105**: β -secondary hydroxyl, Δ 12,13- trisubstituted double bond and 17-carboxylic group, placed within triterpene framework, have been re-



Scheme 3. Bismuth triflate-catalyzed Wagner-Meerwin rearrangement from lupanes do oleanene core.

peatedly exploited for standard synthetic transformations. Thus β -ol secondary hydroxyl group is a key point for glycosylations and/or acylations, while after oxidation to ketone it activates vicinal position for further C-X type functionalization. Double bond serves for direct additions (hydroxylation, epoxidation) as well as for an activation of the allylic positions. Finally, carboxylic group is utilized for both: protective esterification or active ester or amide bond type derivatization. In particular, triterpenoid carboxylic acids have been used as scaffolds in multidirectional syntheses on solid support, with use of specially designed linkers (Wang & Fang, 2011). It should be mentioned, however, that common functionalities placed within PTT framework can feature some unexpected reactivity characteristic. While β -ol susceptibility towards acylating, alkylating and oxidating reagents corresponds well with typical cyclohexanol substrate, the remaining functions exhibit reduced reactivity reflecting steric constraint of their vicinity. The carboxylic group can be converted into esters by action of diazomethane or carboxylate salts alkylation by electrophilic halogenalkanes, but esters thus formed do not undergo hydrolysis under typical basic conditions. The Δ -11,12 double bond is not saturable during catalytic hydrogenation and it is not cleaved during ozonolysis. Additionally, action of other oxidative reagents result in introduction of C-O function into protected **105** (CrO_3 affords unsaturated 11-oxo product and peroxyacetic acid gives 12-ketone) (Farina *et al.*, 1998). These limitations did not seriously hampered a program of semisynthesis from available substrates such as **105** (and isomeric ursolic acid), which brought into being hundreds of new chemical entities, and shed light on PTT potential in drug discovery, during pharmacological studies carried out in last decades.

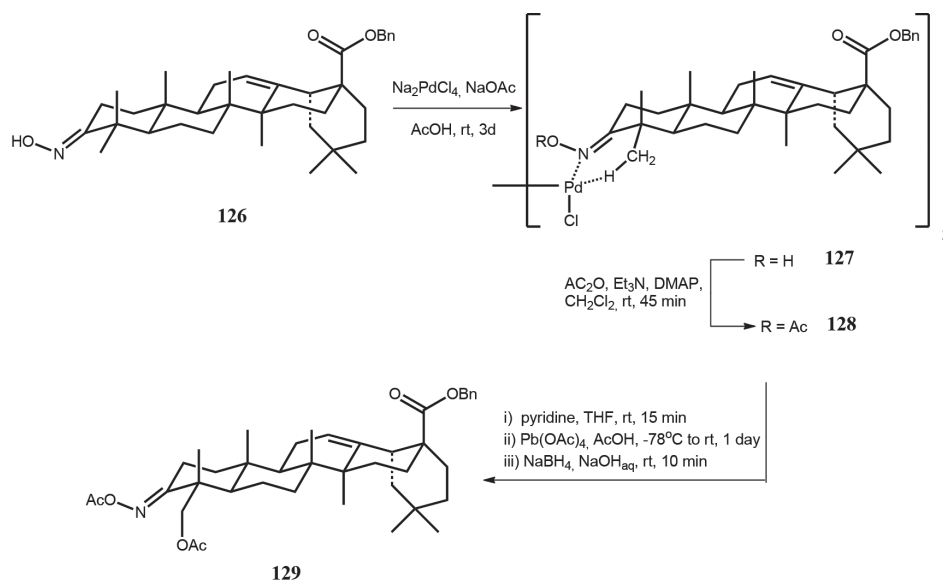
Discussing **105** as starting material for exploratory syntheses should be preceded by information of its role in natural TPP preparation. Conversion to erythrodiol, which involves only one functional group transformation (reduction of C-28 carboxylic function to corresponding primary alcohol) may be considered trivial, preparation of maslinic acid requires, beside carboxylic protection selective hydroxylation at C-2, but elaboration of haptic or myriceric acid. A syntheses proved much more chal-

lenging, because it required chemical oxidative transformation of a specific methyl group. One successful solution of such problem, namely Baldwin's catalytic activation of C-23 methyl group (Baldwin *et al.*, 1985; Neufeldt & Sanford, 2010) is depicted on Scheme 4 below.

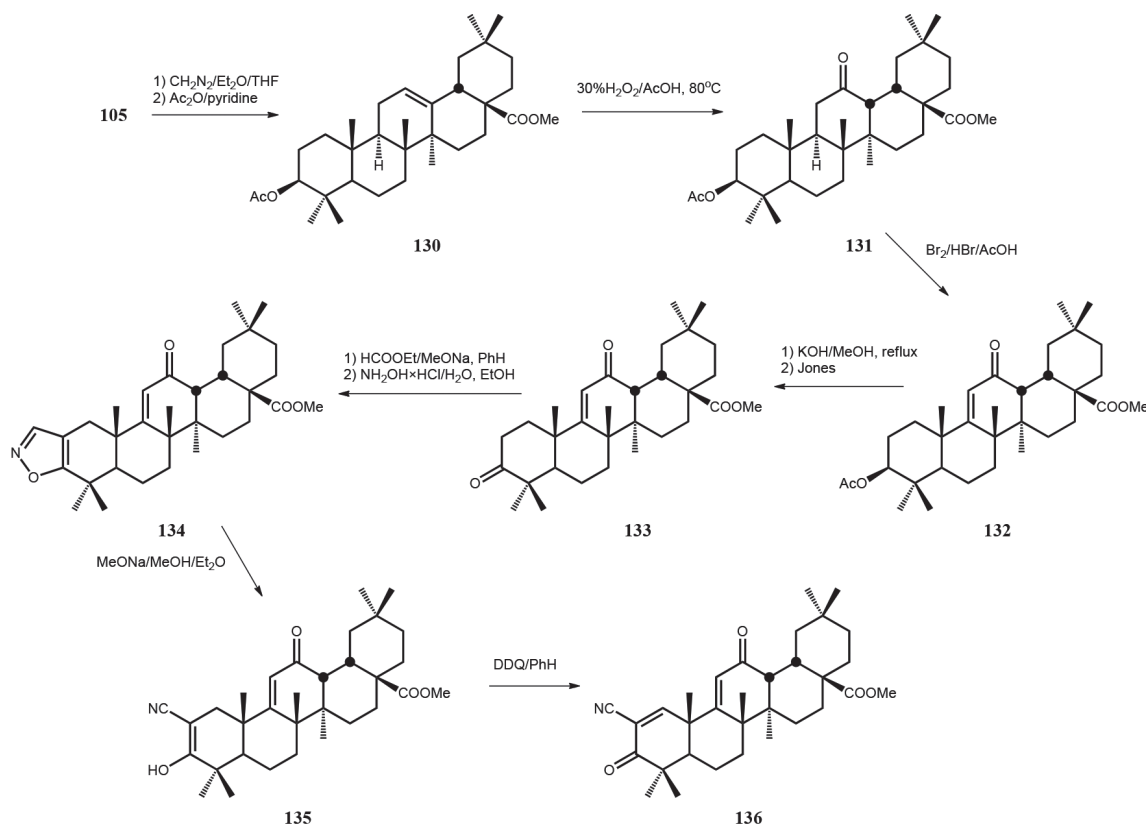
Carboxylic groups of acidic PTT have been functionalized in many ways. When not designated for engagement in an anomeric esterification as in desmosidic saponins, it could be protected as alkyl or benzyl ester in O-alkylation protocol, to facilitate selective transformation of other functionalities. Formation of an amide bond was frequently applied for conjugation with pharmacologically desirable substituents. In a project aimed at novel cytotoxic agents, **105** was conjugated with biologically active natural product — dehydrozingerone, which involved a phenol group esterification, usually considered very susceptible to chemical as well as biological degradation (Tatsuzaki *et al.*, 2007). Application of esterification procedure involving EDCI (1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride) in the presence of DMAP (4-dimethylaminopyridine) resulted in formation of several suitably stable PTT esters which have shown significant cytotoxic effects.

The leading line of chemical modifications of **105** which started with introduction of enone system into ring A, continued with repetition of the transformations in ring C, extension of the conjugated bond system and additional derivatizations. Typical reaction sequence involved conversion of a secondary hydroxyl group into ketone, followed by vicinal substitution-elimination reaction carried out by halogenating or seleno-organic reagents (Sporn *et al.*, 2011). Syntheses, which were guided by inhibition of iNOS test, further led to C-2 formylation with ethyl formate, followed by oxazole ring closure with hydroxylamine and subsequent ring fission with sodium methoxide, which left C-2 cyano substituent allowing for crucial modification of 2-ene-3-one arrangement, present in the original lead compound TP-46 (CDDO) (Konopleva *et al.*, 2004; Sporn *et al.*, 2011; Liby & Sporn, 2012).

Drug candidate TP-151 (2-cyano-3,12-dioxooleana-1,9(11)-dien-28-oic acid; CDDO) selected during the extensive synthetic program carried out by MB Sporn and collaborators, was recognized as a selective Michael



Scheme 4. Catalytic activation and formal hydroxylation of C-23 methyl group.



Black dots represent β -positioned hydrogen atoms, thus indicating stereochemistry of the ring junctions.

Scheme 5. Synthesis of the drug candidate CDDO from protected oleanolic acid (**105**).

acceptor covalently but reversibly interacting with endogenous S- nucleophiles such as glutathione or cysteine residues, although no isolable adducts were obtained in chemical experiments. The original idea to find PTT based inhibitor of inflammatory process was highly successful, providing effective (low nanomolar) inhibitor of COX-2 and inflammatory cytokins. Under name bardoxolone methyl (**XX**) it started, under auspices of Reata Pharmaceuticals (Dallas, TX, USA) parallel clinical trials, first to assess its anticancer properties and later towards advanced chronic kidney disease in patients with type 2 diabetes (Pergola *et al.*, 2011). In parallel with Phase 1 clinical trials **XX** synthesis was elaborated into five-step scalable process, proceeding in about 50% overall yield from **105** (Fu & Gribble, 2013). Evident antioxidant and anti-inflammatory action, believed to operate via induction of transcription factor Nrf2, which reduces activity of the IKK β /NF κ B pathway suggested possibility of various other therapeutic indications. Eventually bardoxolone BEACON clinical study was terminated for safety concerns (Tayek & Kalantar-Zadeh, 2013).

Thus agent CDDO Me has failed in the most important indication, despite shining performance (results) in preclinical study, but inspired many developmental activities, which may prove very important for next PTT drug candidates, which are likely to come along soon. Initial problems with synthesis and formulation were solved, through persistent and effective work towards new pro-drug versions (e.g. C-28 imidazolide and analogs) of the active substance, which offers promise of new generation API. Process chemistry for semi-synthesis for

CDDO analogs is clearly in sight (Fu & Gribble, 2013), which may be taken as an encouragement to start SAR supported exploratory chemistry with numerous polyhydroxyl analogs of **105**.

Among many other attempts to exploit **105** as a lead for new drugs synthetic motifs of lactonization should be mentioned. It has been noticed that its treatment with ozone did not cleave ring C double bond as expected. Instead, additional ring was formed, linking carbons 13 and 28 by ester linkage (Sun *et al.*, 2006). Similar lactonization is observed when **105** is treated with bromine in acetic acid solution. Concomitant introduction of a substituent at C-12 can be handy for further functionalization (Pollier & Goossens, 2012). Another type of lactones can be formed by Bayer-Villiger oxidation of cyclic ketones. This transformation has been exploited for the ring A cleavage and examination of biological activity of 3,4-seco structures derived from PTT *via* two-step oxidation of 3β alcohols (Maitraie *et al.*, 2009). Possibility of a regioselective exocyclic C-2 chain formation, started by base catalysed formylation seems to have much wider scope than thus far demonstrated formation of A-fused heterocyclic ring formation (Chen *et al.*, 2008). Many 3-O-esters of **105**, designed as better soluble and more available congeners have been prepared and tested for biological activity (Pollier & Goossens, 2012). Recently, another example of **105** based SAR and development of its derivative as a new lead compound have been described. A Chinese group, working towards development of new HCV entry inhibitors has found that carboxylic group of **105** and its C-16 vicinity are the modifiable and structure sensitive sites, while the rest of molecule

should be conserved to sustain antiviral activity (Yu *et al.*, 2013). In summary amount of synthetic studies performed on oleanolic acid and its derivatives greatly outnumbered reactions carried out on neutral polyhydroxylic oleanenes isolated directly from plants or obtained from saponins by their controlled degradation.

CONCLUSIONS

Oleanene class of naturally occurring PTT contains some hundreds of individual chemicals (giving rise to much larger set of natural secondary derivatives through biogenetic diversification caused by action of acylating enzymes and glycosyltransferases; similarly, application of synthetic chemistry can easily multiply any subgroup of natural PTT), which should be treated as an invaluable renewable resource of structural diversity, with multiple prospective applications, primary in the human healthcare. Selected 120 PTT structures (1–120) presented in this review, represent a unique set of closely related compounds with high affinity to biopolymeric assembly performing cell biochemistry processes. Split into approximately even collections of neutral and acidic saponin, this arbitrary selected sub-library of triterpenoid plant metabolites provides representative overview of structures and chemotypes encountered in Nature, thus far only partially explored for prospective lead compounds. The main lesson learned from biological activity studies of natural and semi-synthetic oleanenes unanimously prove that the compounds function well in biological environment (in other words: they are inherently biocompatible) exhibiting multitarget functional interactions in micro- and even millimolar concentrations. These compounds are characterized by multitarget activities, which are in principle tunable by chemical modifications. This characteristic suits well novel ideas in drug discovery, which challenge current paradigm, focusing on one molecular target approach (Medina-Franco *et al.*, 2013). Simple chemical concepts, like utility of a Michael acceptor element (represented in Nature within glycyrrhetic acids), has led to useful development in both: PTT semisynthesis and pharmacology. It seems fair to speculate that idea of reversible binding of reactive protein nucleophiles (e.g. cysteine -SH groups) is generally sound, but extended molecular framework (preferably polycyclic) and supplementary functional groups (-OH, esters, ketones, carboxyls, amide etc.) are needed to make it properly selective for expected pharmacological effect. Thus, described oleanene family of PTT represent natural library apparently rich in prospective lead compounds.

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