

Regular paper

Polarization angle dependence of stark absorption spectra of spirilloxanthin bound to the reconstituted LH1 complexes using LH1-subunits isolated from the purple photosynthetic bacterium *Rhodospirillum rubrum**

Tomoko Horibe^{1,2}, Katsunori Nakagawa^{2,3}, Toshiyuki Kusumoto^{1,2}, Ritsuko Fujii^{2,3}, Richard J. Cogdell⁴, Mamoru Nango^{2,3} and Hideki Hashimoto^{1,2,3^{III}}

¹Department of Physics, Graduate School of Science, Osaka City University, Sugimoto, Sumiyoshi-ku, Osaka, Japan; ²CREST/JST, Honcho Kawaguchi, Saitama, Japan; ³The OCU Advanced Research Institute for Natural Science and Technology (OCARINA), Osaka City University, Sugimoto, Sumiyoshi-ku, Osaka, Japan; ⁴Department of Life and Materials Engineering, Graduate School of Engineering, Nagoya Institute of Technology, Gokiso-cho, Showa-ku, Nagoya, Japan; ³Glasgow Biomedical Research Centre, Faculty of Biomedical and Life Sciences, University of Glasgow, Glasgow, UK

Reconstituted LH1 complexes were prepared using the LH1 subunit-type complexes, isolated from the purple photosynthetic bacterium Rhodospirillum (Rs.) rubrum, and purified all-trans spirilloxanthin. Stark absorption spectra of spirilloxanthin bound to both the native and reconstituted LH1 complexes were compared in different polarization angles (x) against the external electric field. From the polarization angle dependence of the Stark absorption spectra, two angles were determined in reference to the direction of transition dipole moment (m) of spirilloxanthin: one is the change in polarizability upon photoexcitation (Δa), $\theta_{\Delta a}$ and the other is the change in static dipole moment upon photoexcitation $(\Delta \mu)$, $\theta_{\Delta \mu}$. Despite the symmetric molecular structure of all-trans spirilloxanthin, its Stark absorption spectra show pronounced values of $\Delta \mu$. This large $\Delta \mu$ values essentially caused by the effect of induced dipole moment through $\Delta \alpha$ both in the cases for native and reconstituted LH1 complexes. However, slightly different values of θ_{Aa} and $\theta_{\Delta u}$ observed for the native LH1 complex suggest that spirilloxanthin is asymmetrically distorted when bound to the native LH1 complex and gives rise to intrinsic $\Delta \mu$ value.

Key words: Stark absorption spectroscopy, polarization angle dependence, LH1 complex, reconstitution, *Rhodospirillum rubrum*, carotenoid, spirilloxanthin

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INTRODUCTION

The photosynthetic light-harvesting (LH) antenna absorbs light energy and transfers the captured excitation energy with high efficiency to the reaction centre (RC) (Ke, 2001). The LH system in photosynthetic purple bacteria generally consists of two types of antenna complexes, LH1 and LH2 (Cogdell *et al.*, 2008). They are both oligomers of a simple monomeric unit consisting of a pair of low molecular weight, hydrophobic apo-proteins (called α and β). This pair of apo-proteins bind, non-covalently two or three bacteriochlorophyll *a* (BChl *a*) and one carotenoid molecule. Optical properties of the pigments can be evaluated on the basis of the available structural data. From this perspective, LH2 whose structure has been determined in atomic resolution served as a cornerstone for the development of a detailed understanding of structure-function relationships of the pigments in LH system (Cogdell et al., 2006). This is indeed a successful example to show the linkage of the structural biology and optical spectroscopy. However, when the detailed structural information has not yet been available, the optical spectroscopy still can afford the information about the structure of pigments bound to the LH complexes. This is exactly what we have done in this presented study for carotenoid bound to the LH1 complex from a purple photosynthetic bacterium. The only one successful X-ray crystal structure analysis for the LH1 complex has been reported for the LH1-RC complexes from a purple photosynthetic bacterium, Rhodopseudomonas (Rps.) palustris (Roszak et al., 2003). However, unfortunately the electron density that should be assignable for carotenoid has not yet been clarified with the reported 4.8 Å resolution.

In order to draw out meaningful structural information using spectroscopy, we have applied Stark absorption spectroscopy to the carotenopid bound to the reconstituted LH1 complex from a purple photosynthetic bacterium, Rhodospirillum (Rsp.) rubrum. This particular spectroscopy is a well-established method with which to investigate the electrostatic properties of pigments embedded in photosynthetic pigment-protein complexes (Middendorf et al., 1993; Yanagi et al., 2004). Reconstitution of carotenoid into photosynthetic pigment-protein complexes has been proved to be powerful to study their functions (Akahane *et al.*, 2004). This is partially because natural LH complexes always have contamination of the several different kinds of Car that are synthesized along the biosynthetic pathway. Hence, the reconstitution is the one possible way to prepare the LH complexes that have single carotenoid moiety. The LH1 complexes can be fully reconstituted in vitro from its constituent parts, i.e., the pigments (BChl *a* and carotenoid) and, α - and β -apo-proteins (Iida et al., 2005; Cogdell et al., 2006). In the previous study (Nakagawa et al., 2008a), we have successfully reconstituted a series of carotenoids having dif-

e-mail address: hassy@sci.osaka-cu.ac.jp

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Abbreviations: BChl, bacteriochlorophyll; LDAO, Lauryldimethylamine-oxide; LH, light-harvesting; PVA, Polyvinyl alcohol; RC, reaction centre; *Rps., Rhodopseudomonas; Rsp., Rhodospirillum*.

ferent extent of conjugation into the LH1 complex of a purple photosynthetic bacterium *Rsp. rubrum*. The reconstituted LH1 complexes were investigated using absorption, fluorescence excitation, and Stark absorption spectroscopies. This study successfully correlates the nonlinear optical parameters determined by Stark absorption spectroscopy with those predicted by semi-empirical molecular orbital calculations and provides us with one of the promising strategies to discuss the conformation of the carotenoid molecules in the LH1 complexes. However, in this previous study, we have supposed that the spirilloxanthin bound to the reconstituted LH1 complex should have symmetric structure. This hypothesis has been tested in this present study.

In this study, we go one step further for the usage of the Stark absorption spectroscopy. The Stark absorption spectra were recorded in different polarization angles (χ) against the external electric field. From this set of measurements, we could determine the angles of both the change in polarizability upon photoexcitation ($\Delta \alpha$) and the change in static dipole moment upon photoexcitation ($\Delta \mu$) with respect to the transition dipole moment (*m*) of carotenoid. Although the chemical structure of all-*trans* spirilloxanthin shown in Fig. 1 is symmetric, its symmetry breaking upon binding to the apo-protein has been extensively studied.



Figure 1. The Chemical structure of all-trans spirilloxanthin.

MATERIALS AND METHODS

Sample preparation. Cultivation and harvesting of the cells of a purple photosynthetic bacterium, *Rsp. rubrum* wild-type strain S1, were performed as previously described (Nakagawa *et al.*, 2008b; 2009). All-*trans*-spirilloxanthin, the LH1 subunit-type complexes (B820 complexes) and the native LH1 complexes were isolated from *Rsp. rubrum* S1 and purified as reported previously (Nakagawa *et al.*, 2008a, 2009). Reconstitution (re-association) of the LH1 complexes (REC_{spx}) were performed using the LH1 subunit-type complex and the purified all-*trans* spirilloxanthin as reported previously (Nakagawa *et al.*, 2008a,b).

The LH1 complexes thus obtained were dispersed in 20 mM Tris. Cl (pH 8.0) buffer solution containing 0.01% lauryldimethylamine-oxide (LDAO), and then uniformly dispersed into polyvinyl alcohol (PVA) film as reported previously (Nakagawa *et al.*, 2008a; 2008b). Briefly, 50 mg of PVA was added in 1 ml solution of LH1 at the concentration of OD_{880nm} about 10, and stirred overnight at 4°C to dissolve PVA. The solution was dropped onto the surface of gold electrodes vapor-deposited on the glass surface and dry up under reduced pressure.

Spectroscopic measurements. Absorption spectra were recorded using a commercially available UV-VIS spectrophotometer (V-670, JASCO) as well as the homebuilt set-up for Stark absorption measurements described below.

Figure 2 shows the block diagram of the home-built set-up for Stark absorption measurements. The details of this set-up have already been reported (Yanagi *et al.*, 2004). For the measurements of steady-state absorption spectra using the Stark set-up, a xenon lamp (150W, L2274, Hamamatsu) was replaced by a tungsten-halogen lamp (55W, H3-12V55W, Phoenix) as a light source of white continuum. The light output from the monochromator (SpectraPro 150, Acton Research) passes through



Figure 2. The Block diagram of the set-up for Stark absorption spectroscopy.

a quartz polarizer (Gran-Thomson prism) to make linear polarization. Direction of the polarization was chosen by rotating the polarizer, and hence the polarization angles against the external electric field, χ , was controlled. Stark absorption spectra were recorded and analyzed in the spectral region from 480 nm to 580 nm (carotenoid absorption region). All the measurements were performed at room temperature.

Analysis of Stark absorption spectra. The theoretical representation of the Stark absorption spectrum was propounded by Liptay (1974). The Stark absorption spectrum ($\Delta A(v)$) can well be reproduced using the 0th, 1st and 2nd order derivatives of the steady-state absorption spectrum (A(v)) as shown in equation (1).

$$\Delta \mathbf{A}(\mathbf{v}) = \left[\mathbf{A}_{\chi} \cdot \mathbf{A}(\mathbf{v}) + \mathbf{B}_{\chi} \cdot \frac{\mathbf{v}}{15hc} \frac{d\left(\mathbf{A}(\mathbf{v}) \right)}{d\mathbf{v}} + \mathbf{C}_{\chi} \cdot \frac{\mathbf{v}}{30h^{2}c^{2}} \frac{d^{2} \left(\mathbf{A}(\mathbf{v}) \right)}{d\mathbf{v}^{2}} \right] \cdot \left| f \cdot \mathbf{E}_{\mathsf{ext}} \right|^{2}$$
(1)

where v represents the frequency of the incident radiation, f is the correction factor for the local field, E_{ext} is the externally applied electric field. The constants, A_{χ} , B_{χ} , C_{χ} , can be determined by spectral fitting. Assuming the contribution of the A_c term is negligible and this is mostly hold true for the LH complexes (Yanagi *et al.*, 2004), the coefficients B_{χ} and C_{χ} can be written as equations (2) and (3).

$$B_{\chi} \approx \frac{2}{5} \operatorname{Tr}(\Delta \alpha) + (3\cos^2 \chi - l) \left(\frac{3}{2} \mathbf{m} \cdot \Delta \alpha \cdot \mathbf{m} - \operatorname{Tr}(\Delta \alpha)\right)$$
(2)

$$C_{\chi} = 5 |\Delta \mu|^2 + (3\cos^2 \chi - 1)(3(m \cdot \Delta \mu)^2 - |\Delta \mu|^2)$$
(3)

Here *m* shows the unit vector of the transition dipole moment. Nonlinear optical parameters, $m \cdot \Box \alpha \cdot m$, $Tr(\Box \alpha)$, $m \cdot \Box \mu$ and $|\Box \mu|$, can be determined by the fitting of the curves drawn by equations (2) and (3) to the experimentally determined values of B_{χ} and C_{χ} for $\chi = 0^{\circ}$, 30° , 54.7° , 70° and 90° . The angles of $\Box \alpha$ and $\Box \mu$ with respect to the direction of m, $\theta_{\Delta \alpha}$ and $\theta_{\Delta \mu}$, respectively, can be determined by the following equations.

$$\frac{\mathbf{m} \cdot \Delta \mathbf{a} \cdot \mathbf{m}}{\mathrm{Tr}(\Delta \mathbf{a})} \approx \cos \theta_{A \mathbf{a}} \tag{4}$$

$$\frac{\boldsymbol{m} \cdot \Delta \boldsymbol{\mu}}{|\Delta \boldsymbol{\mu}|} = \cos \theta_{\Delta \boldsymbol{\mu}} \tag{5}$$



Figure 3. Absorption spectra of the native (solid line) and reconstituted (REC_{SPXY} dotted line) LH1 complexes in PVA films at room temperature.

RESULTS AND DISCUSSION

Steady-state absorption spectra

Figure 3 shows the steady-state absorption spectra of the native and reconstituted (REC_{spx}) LH1 complexes in the PVA films. Both spectra show basically the same profiles. Characteristic absorption band from 450 to 470 nm can be assigned to the $S_2 \leftarrow S_0$ transition of spirilloxanthin bound to the LH1 complexes. The bands peaking at 377, 580, and 878 nm were assigned to the Soret, Q_x and Q_y bands of BChl *a*, respectively.

Stark absorption spectra

Figure 4 shows the Stark absorption spectra at $\chi = 54.7^{\circ}$ (open circles) and the results of fittings (thin lines) for (a) native and (b) reconstituted LH1 complexes in PVA films. The 0th, 1st and 2nd order derivatives of the steady-state absorption spectra that contribute to the fittings were also shown in the Figure. The Stark absorption spectra were well reproduced by these fittings. The 0th derivatives contribute relatively small and negatively to the Stark absorption spectra. This supports the assumption for equations (2) and (3), where the transition dipole-moment polarizability assumed to be negligible (Bublitz *et al.* 1997b).

Figure 5 shows the plots of B_{χ} and C_{χ} determined by the fittings using equation (1) against χ for the native (open triangles) and reconstituted (open circles) LH1 complexes in PVA films. The results of the fittings using equations (2) and (3) were also shown in solid lines. Table 1 summarizes the nonlinear optical parameters determined in this study. Small difference in the values of Tr($\Box \alpha$), between the native and reconstituted LH1 complexes should be converged after iterate experiments.

A molecule having strict center of symmetry, like alltrans spirilloxanthin, should not have any values of static dipole moment (μ_i). Therefore in this case the change in the static dipole moment ($\Delta \mu$) should solely come from the contribution of the induced dipole moment ($\Delta \alpha \cdot E$). Carotenoid molecules typically show large $\Delta \mu$ values (Gottfried *et al.*, 1991; Krawczyk *et al.*, 1995). This is thought to be due to the contribution of the induced



Figure 4. Stark absorption spectra at $\chi = 54.7^{\circ}$ (open circles) (a) native and (b) reconstituted LH1 complexes in PVA films with the results of spectral fitting (thin solid lines). Spectra responsible for the $S_2 \leftarrow S_0$ absorption region of spirilloxanthin are shown. The contributions of the 0th (dotted lines), 1st (broken lines) and 2nd (thick solid lines) order derivatives of the original absorption spectra are also shown for comparison.

However, the native LH1 complex shows slight but different values of $\theta_{\perp\mu}$ and $\theta_{\perp\mu}$ even though the experimental errors taken into consideration (see Table 1). This is a marked and new finding that relates to discuss the molecular structure of carotenoid bound to the native LH1 complexes. The cause of the different $\theta_{\perp\mu}$ and $\theta_{\perp\mu}$ values is thought to be due to the contribution of the μ_i term in equation (6). The μ_i value can be generated when the molecule loses its center of symmetry. Therefore, the present finding suggests that the molecular structure of all-*trans* spirilloxanthin is somehow deformed so as to lose its center of symmetry according to pigment-protein interaction when bound to the native LH1 complexes.

Table 1. Nonlinear optical parameters determined for the native
and reconstituted LH1 complexes in PVA.
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The parameters were determined from the fittings of Stark absorption spectra observed in different χ values.

Carotenoid	Native	RECspx
Tr(Δα)[ų/f³]	1700±200	1480±200
Δμ [D/f]	7.7±0.5	7.0±0.6
$\theta_{\Delta a}[\circ]$	28±2	30±1
$\theta_{\Delta\mu}[\circ]$	36±2	30±1



CONCLUSION

The polarization angle dependence of the Stark absorption spectra of all-*trans* spirilloxanthin bound to the native and reconstituted LH1 complexes have been investigated. The large $\Box \mu$ value of spirilloxanthin bound to the reconstituted LH1 complex can be explained solely due to the effect of induced dipole moment. This supports the validity of the assumption of the symmetric molecular structure of spirilloxanthin bound to the reconstituted LH1 complex. However, the native LH1 complex shows slightly different values of $\theta_{\perp\mu}$ and $\theta_{\perp\mu}$ and hence suggests the presence of asymmetrical deformation of all-*trans* spirilloxanthin when bound to the native LH1 complex. This present study shows the usefulness of Stark absorption spectroscopy to discuss the molecular structure of carotenoids bound to the pigmentprotein complexes.

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