

*Communication*

**Conformational properties of *N',N'*-dimethylamides of *N*-acetyldehydroalanine and *N*-acetyl-(*Z*)-dehydrophenyl - alanine<sup>★☉</sup>**

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**Conformational preferences of Ac- $\Delta$ Ala-NMe<sub>2</sub> and Ac-(*Z*)- $\Delta$ Phe-NMe<sub>2</sub> were studied and compared with those of their monomethyl counterparts as well as with those of their saturated analogues. X-Ray data and energy calculations revealed a highly conservative conformation of the dehydro dimethylamides, which is located in a high-energy region of the Ramachandran map.**

An essential point in peptide-targeted molecular design is introducing a specialised amino acid into an elaborated molecule, which confers local constraints on the backbone main chain and the side chain moiety. The specialised amino acids are, e.g.,  $\alpha,\beta$ -dehydroamino acids. Among these,  $\Delta$ Ala occurs most often

and (*Z*)- $\Delta$ Phe is the most commonly used modifier. On the other hand, numerous modifications of peptides and peptide mimetics are based on the tertiary amide bond. The latter can be modelled by the *N',N'*-dimethylamide group. Within the Ac- $\Delta$ Ala-NMe<sub>2</sub> and Ac-(*Z*)- $\Delta$ Phe-NMe<sub>2</sub> molecules, we combined the C <sup>$\alpha$</sup>  =

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**Abbreviation:**  $\Delta$ , dehydro.

$C^\beta$  double bond and the  $N',N'$ -dimethylamide group, and investigated conformations of these compounds in the solid state and in the gas phase. Ac- $\Delta$ Ala-NHMe and Ac-L-Ala-NMe<sub>2</sub> as well as Ac-(Z)- $\Delta$ Phe-NHMe and Ac-L-Phe-NMe<sub>2</sub> were taken for comparison.

## MATERIALS AND METHODS

Ac- $\Delta$ Ala-NMe<sub>2</sub>, Ac-L-Ala-NMe<sub>2</sub>, Ac-(Z)- $\Delta$ Phe-NMe<sub>2</sub> and Ac-L-Phe-NMe<sub>2</sub> were synthesised using standard methods. Their structures were solved by X-ray analysis. X-Ray data for Ac- $\Delta$ Ala-NHMe were taken from [1] and those for Ac-(Z)- $\Delta$ Phe-NHMe from [2]. The AM1 and DFT/B3LYP/6-31G\*\* methods were used for theoretical calculations in the gas phase.

## RESULTS AND DISCUSSION

Figure 1 A–F presents the molecular crystal conformations of Ac- $\Delta$ Ala-NMe<sub>2</sub>, Ac- $\Delta$ Ala-NHMe, Ac-L-Ala-NMe<sub>2</sub>, Ac-(Z)- $\Delta$ Phe-NMe<sub>2</sub>,

Ac-(Z)- $\Delta$ Phe-NHMe and Ac-L-Phe-NMe<sub>2</sub> in the solid state and those of all their low-energy conformers in the gas phase.

In crystal, Ac- $\Delta$ Ala-NMe<sub>2</sub> adopts  $\phi, \psi = -43^\circ, 127^\circ$  and Ac-L-Ala-NMe<sub>2</sub> adopts  $\phi, \psi = -82^\circ, 157^\circ$ . Two independent molecules with  $\phi, \psi = -172^\circ, 154^\circ$  and  $163^\circ, -169^\circ$  were found for Ac- $\Delta$ Ala-NHMe [1]. As can be seen, the conformations of these three compounds differ significantly from each other, and that of Ac- $\Delta$ Ala-NMe<sub>2</sub> is located in the high-energy region (region H [3]) of the Ramachandran diagram. Moreover, the out-of-plane parameters,  $\chi_N$  and  $\chi_C$ , and the twisting parameter  $\tau$  [4], of the C-terminal amide bond of Ac- $\Delta$ Ala-NMe<sub>2</sub> indicate a considerably high degree of non-planarity, whereas this moiety both in Ac-L-Ala-NMe<sub>2</sub> and Ac- $\Delta$ Ala-NHMe is rather flat.

The conformational map of Ac- $\Delta$ Ala-NMe<sub>2</sub> (Fig. 2A) has two minima and their mirror image. The energy difference and the magnitude of the separation between them amount to 4.2 and 6.3 kcal mol<sup>-1</sup>, respectively. The lower en-

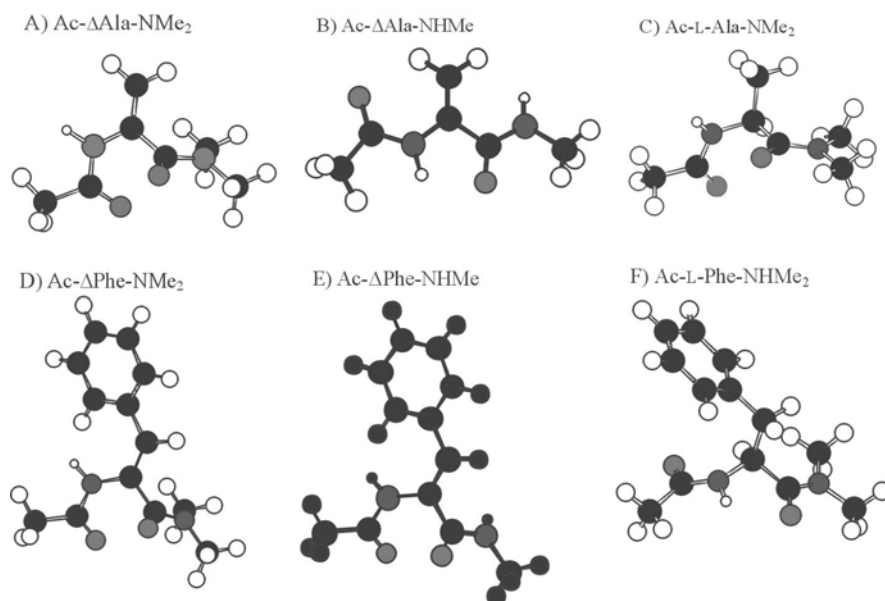
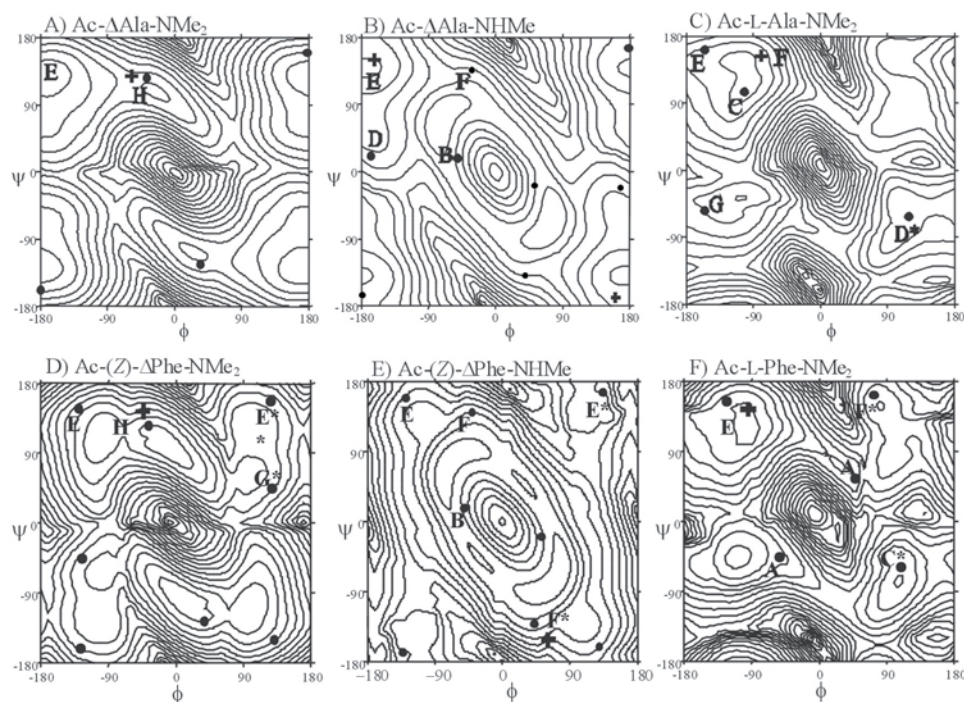


Figure 1. The molecular conformations of the studied molecules in the solid state.

Ac-(Z)- $\Delta$ Phe-NHMe and Ac-L-Phe-NMe<sub>2</sub>. Figure 2 A–F gives their AM1 energy maps, in the  $\phi, \psi$  torsion space *in vacuo*, together with the DFT/B3LYP/6-31G\*\* re-optimised min-

ergy minimum ( $\phi, \psi = 179^\circ, 154^\circ$ ) is positioned in the region of extended conformations. The other ( $\phi, \psi = -37^\circ, 124^\circ$ ) is located in a narrow and relatively highly separated area in region



**Figure 2.** Energy maps in the  $\phi$ - $\psi$  torsion space *in vacuo* computed by the AM1 method on a grid with  $10^\circ$  spacing.

The isopotential lines are drawn every  $1 \text{ kcal mol}^{-1}$ . +, Crystal structures; •, DFT/B3LYP/6-31G\*\* minima.

H of the Ramachandran diagram. The crystal conformer ( $\phi, \psi = -42^\circ, 127^\circ$ ) corresponds exactly to the second minimum. All these data taken together indicate some limited conformational freedom of the Ac- $\Delta$ Ala-NMe<sub>2</sub> molecule. Both the conformation of Ac- $\Delta$ Ala-NMe<sub>2</sub> in crystal and those calculated display a significant deformation of the tertiary amide bond, which means that this non-planarity is an intrinsic feature of the Ac- $\Delta$ Ala-NMe<sub>2</sub> molecule.

The conformational map of Ac- $\Delta$ Ala-NHMe (Fig. 2B) has four minima (and their mirror image). The global minimum ( $\phi, \psi = -179^\circ, 166^\circ$ ), located in region E, is populated in the crystal structure. The remainder, one conformer in the moderate energy bridge region (region B) ( $\phi, \psi = -52^\circ, 20^\circ$ ) and the others in region D ( $\phi, \psi = -168^\circ, 22^\circ$ ) and F ( $\phi, \psi = -42^\circ, 138^\circ$ ) have much higher energies ( $\Delta E = 5.1, 5.8, \text{ and } 7.3 \text{ kcal mol}^{-1}$ , respectively). As is the case for Ac- $\Delta$ Ala-NMe<sub>2</sub>, the C-terminal amide bond reveals some deformation, in all the con-

formers. The deformation, however, is smaller, particularly that characterised by the out-of-plane parameter  $\chi_C$  and the twisting parameter  $\tau$ .

The Ac-L-Ala-NMe<sub>2</sub> molecule in the gas phase (Fig. 2C) has four minima, which in the order of increasing energy are as follows: in region E ( $\phi, \psi = -157^\circ, 161^\circ$ ), in region C ( $\phi, \psi = -101^\circ, 109^\circ$ ), in region G ( $\phi, \psi = -160^\circ, -47^\circ$ ), and in region D\* ( $\phi, \psi = 114^\circ, -59^\circ$ ). The energy difference and the barrier between the two former minima equal to only  $1.0 \text{ kcal mol}^{-1}$  and  $1.6 \text{ kcal mol}^{-1}$ , respectively. The conformation found in the crystal structure lies on this flat potential energy surface. So, the molecule has a high degree of conformational freedom. The out-of-plane parameters of the tertiary amide in all the conformers except conformer G point to a fairly planar system.

In the solid state, Ac-(Z)- $\Delta$ Phe-NMe<sub>2</sub> adopts  $\phi, \psi = -50^\circ, 133^\circ$  whereas Ac-(Z)- $\Delta$ Phe-NHMe

**Table 1. Selected conformational parameters of six sample molecules in the solid state and of all their low-energy conformers in gas phase.**

The conformers are labelled according to the name of the Ramachandran map regions [3].

Compound	Conformer	Energy difference kcal mol <sup>-1</sup>	$\phi$	$\psi$	$\chi^2$	$\chi_C$	$\chi_N$	$\tau$
Ac- $\Delta$ Ala-NMe <sub>2</sub>	(H) <sup>a</sup>		-43.1	127.9	-	-5.5	-11.0	171.6
	E	0.0	179.0	153.9	-	-2.1	-16.7	164.0
	H	4.2	-37.3	123.8	-	-6.6	-26.2	168.0
Ac- $\Delta$ Ala-NHMe	(E) <sup>a,b</sup>		-172.2	154.1	-	-1.0	-2.0	-179.1
	(E*) <sup>a,b</sup>		162.6	-169.4	-	1.4	0.2	-178.6
	E	0.0	178.9	166.4	-	0.2	-11.4	175.0
	B	5.1	-52.3	19.8	-	1.2	21.3	-172.3
	D	5.8	-167.8	22.4	-	1.1	30.1	-171.0
	F	7.3	-42.0	138.0	-	-4.9	-22.3	174.1
Ac-L-Ala-NMe <sub>2</sub>	(F) <sup>a</sup>		-81.8	157.4	-	-3.3	-1.4	178.7
	E	0.0	-156.6	161.0	-	-6.7	-3.9	177.0
	C	1.0	-101.2	109.3	-	0.5	0.0	175.3
	G	7.5	-160.0	-47.5	-	-1.5	-20.5	170.8
	D*	8.2	113.9	-59.5	-	-0.3	-1.7	178.0
Ac- $\Delta$ Phe-NMe <sub>2</sub>	(F) <sup>a</sup>		-49.5	132.6	26.4	-3.2	-7.9	170.7
	H	0.00	-37.1	126.4	-29.7	-6.4	-26.2	167.6
	E	1.05	-130.4	148.0	16.5	-2.5	-17.8	163.8
	E*	3.11	126.4	157.0	-19.6	-1.3	-22.3	160.3
	G*	6.97	128.4	45.7	-17.7	3.0	31.2	-160.0
Ac- $\Delta$ Phe-NHMe	(F*) <sup>a,c</sup>		56.8	-148.7	39.8	1.9	6.8	-174.6
	B	0.00	-51.0	19.1	-31.5	1.3	22.5	-172.0
	E	2.12	-129.8	160.6	18.5	-1.2	-19.5	172.5
	E*	2.31	133.1	168.2	-19.6	-1.4	19.6	172.7
	F	2.79	-41.7	140.5	-27.2	-4.5	-24.8	173.4
Ac-L-Phe-NMe <sub>2</sub>	(F) <sup>a</sup>		-99.5	148.5	-85.6	-3.5	3.6	179.9
	E	0.00	-125.6	154.4	-85.9	-3.1	-0.3	177.3
	A*	4.84	47.1	56.2	-79.0	-5.4	-8.1	-176.3
	A	5.37	-53.4	-44.1	-72.8	6.9	9.4	179.3
	F*	6.04	71.4	163.2	-87.1	-0.3	-2.9	174.2
	C*	6.31	109.1	-59.7	-75.1	0.6	-3.2	176.9

<sup>a</sup> Crystal data values; <sup>b</sup> Ref. [1]; <sup>c</sup> Ref. [2]

and Ac-L-Phe-NMe<sub>2</sub> assume  $\phi, \psi = 57^\circ, -149^\circ$  and  $\phi, \psi = -100^\circ, 149^\circ$ , respectively. As can be seen, the conformations of the dehydro compounds are similar and that of the saturated dimethylamide is quite different. Out of plane parameters point to the most non-planar tertiary amide bond in the Ac-(Z)- $\Delta$ Phe-NMe<sub>2</sub> molecule and the almost planar tertiary amide bond in Ac-L-Phe-NMe<sub>2</sub>.

The conformational map of Ac-(Z)- $\Delta$ Phe-NMe<sub>2</sub> (Fig. 2D) has four minima and their mirror image. The global minimum ( $\phi, \psi = -37^\circ, 126^\circ$ ) is located in an area of a clearly distinct topology, in region H of the Rama-

chandran diagram. The conformation found in the crystal ( $\phi, \psi = -50^\circ, 132^\circ$ ) corresponds quite well to this minimum. The second-lowest minimum ( $\phi, \psi = -130^\circ, 148^\circ$ ), of energy difference 1.1 kcal mol<sup>-1</sup>, is located in the region of C<sub>5</sub> conformations (region E). The remaining two minima occur in the upper-right quarter of the map and have much higher energies ( $\Delta E = 3.1$  and 7.0 kcal mol<sup>-1</sup>). The Ac-(Z)- $\Delta$ Phe-NMe<sub>2</sub> molecule in crystal as well as in all the calculated conformers displays a significant non-planarity of the tertiary amide bond, which means that this non-planarity is its intrinsic feature. In all four theoretical con-

formers as well as in the crystal conformation, the angle  $\chi^2$  is relatively small and does not exceed  $\pm 30^\circ$ . This indicates a  $\pi$ -coupling tendency between the phenyl ring and the C <sup>$\alpha$</sup> =C <sup>$\beta$</sup>  double bond, which provides a strong steric constraint on the backbone torsion  $\phi$  angle and prevents the molecule from assuming the fully extended conformation.

The conformational map of Ac-(Z)- $\Delta$ Phe-NHMe (Fig. 2E) also has four minima (and their mirror image). The lowest ( $\phi, \psi = -51^\circ, 19^\circ$ ) is located in bridge region B. The remainder, in region E ( $\phi, \psi = -130^\circ, 161^\circ$ ), in region E\* ( $\phi, \psi = 133^\circ, 168^\circ$ ), and in region F ( $\phi, \psi = -42^\circ, 141^\circ$ ) ( $\Delta E = 2.1, 2.3, \text{ and } 2.8 \text{ kcal mol}^{-1}$ , respectively). In contrast to Ac-(Z)- $\Delta$ Phe-NMe<sub>2</sub>, the conformation of Ac-(Z)- $\Delta$ Phe-NHMe found in the crystal ( $\phi, \psi = 57^\circ, -149^\circ$ ) corresponds to the highest energy minimum, which is placed in the mirror image region F\*. As is the case for Ac-(Z)- $\Delta$ Phe-NMe<sub>2</sub>, the C-terminal amide bond of all four conformers reveals some deformation, although – particularly for the out-of-plane parameter  $\chi_C$  and the twisting parameter  $\tau$  – smaller. The value of the angle  $\chi^2$  in all the calculated conformers and that in the crystal (about  $\pm 30^\circ$  on average) indicates a  $\pi$ -coupling tendency between the phenyl ring and the C <sup>$\alpha$</sup> =C <sup>$\beta$</sup>  double bond.

Energy calculations show that the Ac-L-Phe-NMe<sub>2</sub> molecule represents an entirely different conformational pattern. The conformational map (Fig. 2F) has five minima, which in the order of increasing energy are as follows: in region E ( $\phi, \psi = -126^\circ, 154^\circ$ ), in region A\* ( $\phi, \psi = 47^\circ, 56^\circ$ ), in region A ( $\phi, \psi = -53^\circ, -44^\circ$ ), in region F\* ( $\phi, \psi = 71^\circ, 163^\circ$ ) and in region C\* ( $\phi, \psi = 109^\circ, -60^\circ$ ). The conformation found in the crystal structure ( $\phi, \psi = -100^\circ, 149^\circ$ ) lies in the broad and flat area surrounding the minimum E. The out-of-plane parameters of the tertiary amide in all conformers point to a rather planar system. The angle  $\chi^2$  is close to  $-90^\circ$  and allows the molecule to

assume the fully extended conformation. All this indicates that the compound has a high degree of conformational freedom.

## CONCLUSION

X-Ray analysis of Ac- $\Delta$ Ala-NMe<sub>2</sub> and Ac-(Z)- $\Delta$ Phe-NMe<sub>2</sub> reveals a highly conservative conformation of these compounds. Both adopt a structure with the torsion angles  $\phi, \psi = \sim -45^\circ, \sim 130^\circ$  and show some non-planarity of its C-terminal amide. This conformation is located in the high-energy region (region H) of Ramachandran diagram, which is not readily accessible to their saturated analogues. Theoretical calculations on isolated Ac- $\Delta$ Ala-NMe<sub>2</sub> and Ac-(Z)- $\Delta$ Phe-NMe<sub>2</sub> molecules corroborate these findings. A comparison of all respective conformational data indicates that Ac- $\Delta$ Ala-NMe<sub>2</sub> and Ac-(Z)- $\Delta$ Phe-NMe<sub>2</sub> share more conformational features with their monomethyl rather than saturated analogues.

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